

Chemistry Study Guide

Periodic Table of the Elements

¹ ₁ H																	² ₂ He																															
³ ₃ Li	⁴ ₄ Be																	⁵ ₅ B	⁶ ₆ C	⁷ ₇ N	⁸ ₈ O	⁹ ₉ F	¹⁰ ₁₀ Ne																									
¹¹ ₁₁ Na	¹² ₁₂ Mg																	¹³ ₁₃ Al	¹⁴ ₁₄ Si	¹⁵ ₁₅ P	¹⁶ ₁₆ S	¹⁷ ₁₇ Cl	¹⁸ ₁₈ Ar																									
¹⁹ ₁₉ K	²⁰ ₂₀ Ca	²¹ ₂₁ Sc	²² ₂₂ Ti	²³ ₂₃ V	²⁴ ₂₄ Cr	²⁵ ₂₅ Mn	²⁶ ₂₆ Fe	²⁷ ₂₇ Co	²⁸ ₂₈ Ni	²⁹ ₂₉ Cu	³⁰ ₃₀ Zn	³¹ ₃₁ Ga	³² ₃₂ Ge	³³ ₃₃ As	³⁴ ₃₄ Se	³⁵ ₃₅ Br	³⁶ ₃₆ Kr																															
³⁷ ₃₇ Rb	³⁸ ₃₈ Sr	³⁹ ₃₉ Y	⁴⁰ ₄₀ Zr	⁴¹ ₄₁ Nb	⁴² ₄₂ Mo	⁴³ ₄₃ Tc	⁴⁴ ₄₄ Ru	⁴⁵ ₄₅ Rh	⁴⁶ ₄₆ Pd	⁴⁷ ₄₇ Ag	⁴⁸ ₄₈ Cd	⁴⁹ ₄₉ In	⁵⁰ ₅₀ Sn	⁵¹ ₅₁ Sb	⁵² ₅₂ Te	⁵³ ₅₃ I	⁵⁴ ₅₄ Xe																															
⁵⁵ ₅₅ Cs	⁵⁶ ₅₆ Ba			⁷² ₇₂ Hf	⁷³ ₇₃ Ta	⁷⁴ ₇₄ W	⁷⁵ ₇₅ Re	⁷⁶ ₇₆ Os	⁷⁷ ₇₇ Ir	⁷⁸ ₇₈ Pt	⁷⁹ ₇₉ Au	⁸⁰ ₈₀ Hg	⁸¹ ₈₁ Tl	⁸² ₈₂ Pb	⁸³ ₈₃ Bi	⁸⁴ ₈₄ Po	⁸⁵ ₈₅ At	⁸⁶ ₈₆ Rn																														
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Objective:

To understand the *structure of an atom*, and the *nature + properties* of subatomic particles ✎

Matter:

Anything that has **mass** and **occupies space** ✎

Examples: air, gases, liquids, solids!

Atoms:

Atoms are the **smallest, indivisible particle** ✎

- **Outer part** → *shell/orbit/energy level* → carries the electrons (e^-)
- **Inner part** → *nucleus* → houses the protons (p^+) and neutrons (n)
- Outer & Inner part → together these are the **subatomic particles**

Subatomic Particle Table:

Particle	Symbol	Mass (amu)	Charge	Location
Proton	p^+	1	+1	Nucleus
Neutron	n	1	0	Nucleus
Electron	e^-	~ 0	-1	Orbit/Shell

Thermodynamics Chemistry Study Guide Answers

Daniel C. Tofan



Thermodynamics Chemistry Study Guide Answers:

Organic Chemistry Study Guide with Solutions Manual Neil E. Schore, 2007 The guide includes chapter introductions that highlight new material chapter outlines detailed comments for each chapter section a glossary and solutions to the end of chapter problems presented in a way that shows students how to reason their way to the answer *Thermodynamics Problem Solving in Physical Chemistry* Kathleen E. Murphy, 2020-03-23 Thermodynamics Problem Solving in Physical Chemistry Study Guide and Map is an innovative and unique workbook that guides physical chemistry students through the decision making process to assess a problem situation create appropriate solutions and gain confidence through practice solving physical chemistry problems The workbook includes six major sections with 20 30 solved problems in each section that span from easy single objective questions to difficult multistep analysis problems Each section of the workbook contains key points that highlight major features of the topic to remind students of what they need to apply to solve problems in the topic area Key Features Provides instructor access to a visual map depicting how all equations used in thermodynamics are connected and how they are derived from the three major energy laws Acts as a guide in deriving the correct solution to a problem Illustrates the questions students should ask themselves about the critical features of the concepts to solve problems in physical chemistry Can be used as a stand alone product for review of Thermodynamics questions for major tests

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