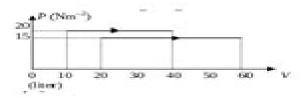
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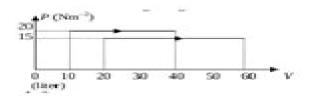


Thermodynamics - problems and solutions

The first law of thermodynamics

 Based on graph P-V below, what is the ratio of the work done by the gas in the process I, to the work done by the gas in the process II?

Known:



Process 1:

Pressure (P) = 20 N/m²

Initial volume (V₁) = 10 liter = 10 dm³ = 10×10^{-3} m³

Final volume $(V_2) = 40 \text{ liter} = 40 \text{ dm}^3 = 40 \times 10^3 \text{ m}^3$

Process 2:

Process (P) = 15 N/m²

Initial volume $(V_1) = 20 \text{ liter} = 20 \text{ dm}^3 = 20 \times 10^{-3} \text{ m}^3$

Final volume $(V_2) = 60 \text{ liter} = 60 \text{ dm}^3 = 60 \times 10^{-3} \text{ m}^3$

Wanted The ratio of the work done by gas

Solution:

Thermodynamics Problems And Solutions

Xiang Xie

Thermodynamics Problems And Solutions:

Unveiling the Magic of Words: A Report on "Thermodynamics Problems And Solutions"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their ability to kindle emotions, provoke contemplation, and ignite transformative change is actually aweinspiring. Enter the realm of "**Thermodynamics Problems And Solutions**," a mesmerizing literary masterpiece penned by a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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