

Pre Lab:

Before you begin the lab, you must calculate the equivalent weight of the 5 ammonium salts listed in table 1. See below for some hints on how to complete the table:

Table 1: Equivalent Weight of the Five Possible Ammonium Salts

Salt	Formula	Molar Mass (g/mol)	Number of equivalents (eq/mol)	Equivalent weight (g/eq)
Ammonium Bromide	NH_4Br	97.94 g/mol	1 eq/mol	97.94 g/eq
Ammonium Chloride	NH_4Cl	53.49 g/mol	1 eq/mol	53.49 g/eq
Ammonium Iodide	NH_4I	144.94 g/mol	1 eq/mol	144.94 g/eq
Ammonium Nitrate	NH_4NO_3	80.05 g/mol	1 eq/mol	80.05 g/eq
Ammonium Sulfate	$(\text{NH}_4)_2\text{SO}_4$	132.15 g/mol	2 eq/mol	66.08 g/eq

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