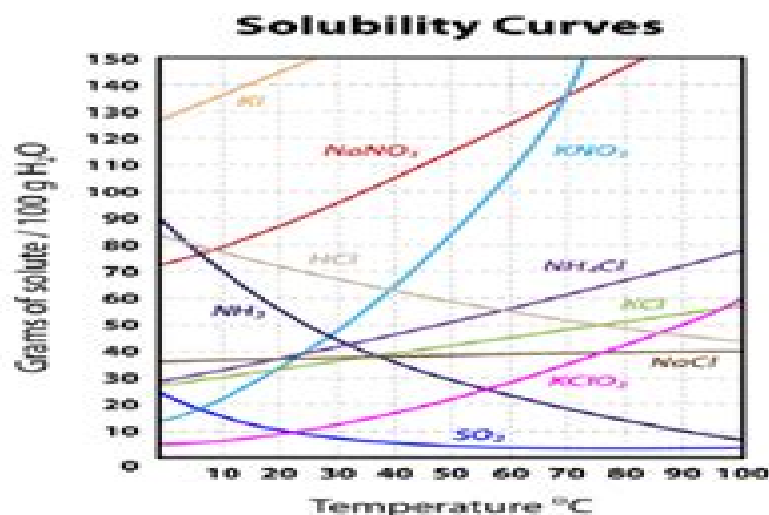


WS 6.1: Interpreting Solubility Curves

Directions: Use the Solubility Curve Table below to answer all questions. So you do not have to keep flipping your paper over you might want to put the solubility curve on your phone or computer screen when answering the questions on the back. All questions assume the amount of water is 100 ml.



- What is the solvent in this Solubility Curve?
water
- Saturation points are given per how much solvent?
100 ml or 100 g
- What is the saturation point of NaNO_3 at 10°C , 40°C , & 80°C ?
@ 10°C = ~79 g
@ 40°C = ~105 g
@ 80°C = ~146 g
- How much NH_3 can you dissolve in water at 10°C , 30°C , & 90°C ?
@ 10°C = ~90 g
@ 30°C = ~45 g
@ 90°C = ~10 g
- What is the maximum amount of KCl that can be dissolved in water at 20°C , 50°C , & 90°C ?
@ 20°C = ~33 g
@ 50°C = ~42 g
@ 90°C = ~53 g
- If you want to make a saturated solution of NaCl how many grams would you dissolve in 100 grams of water @ 80°C ?
About 40 grams
- Except for KCl What substance is the most soluble in water at 20°C ?
 NaNO_3
- Except for KCl what substance is the most soluble in water at 72°C ?
 KNO_3
- Look up The boiling points for SO_2 , HCl , and NH_3 and write them below:
B.P. SO_2 = -10°C
B.P. HCl = -85°C
B.P. NH_3 = -33°C
- Based on the boiling points you looked up, are SO_2 , HCl , & NH_3 solids, liquids, or gases at 0°C ? (and anywhere else on the solubility curve)
gases

Solubility Curves Ws 2 Answers

Didier Musso

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Solubility Curves Ws 2 Answers:

The book delves into Solubility Curves Ws 2 Answers. Solubility Curves Ws 2 Answers is an essential topic that must be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Solubility Curves Ws 2 Answers, encompassing both the fundamentals and more intricate discussions.

1. The book is structured into several chapters, namely:

- Chapter 1: Introduction to Solubility Curves Ws 2 Answers
- Chapter 2: Essential Elements of Solubility Curves Ws 2 Answers
- Chapter 3: Solubility Curves Ws 2 Answers in Everyday Life
- Chapter 4: Solubility Curves Ws 2 Answers in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Solubility Curves Ws 2 Answers. The first chapter will explore what Solubility Curves Ws 2 Answers is, why Solubility Curves Ws 2 Answers is vital, and how to effectively learn about Solubility Curves Ws 2 Answers.
3. In chapter 2, the author will delve into the foundational concepts of Solubility Curves Ws 2 Answers. The second chapter will elucidate the essential principles that must be understood to grasp Solubility Curves Ws 2 Answers in its entirety.
4. In chapter 3, this book will examine the practical applications of Solubility Curves Ws 2 Answers in daily life. The third chapter will showcase real-world examples of how Solubility Curves Ws 2 Answers can be effectively utilized in everyday scenarios.
5. In chapter 4, the author will scrutinize the relevance of Solubility Curves Ws 2 Answers in specific contexts. The fourth chapter will explore how Solubility Curves Ws 2 Answers is applied in specialized fields, such as education, business, and technology.
6. In chapter 5, this book will draw a conclusion about Solubility Curves Ws 2 Answers. This chapter will summarize the key points that have been discussed throughout the book.

This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. This book is highly recommended for anyone seeking to gain a comprehensive understanding of Solubility Curves Ws 2 Answers.

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