

Derive the following equations:

1. Show the following thermodynamic equations:

$$d(\Delta G_{\text{mix}}) = Td(\Delta S_{\text{mix}}) - d(\Delta H_{\text{mix}}) + d(\Delta PV)_{\text{mix}} = -S_{\text{mix}}dT + V_{\text{mix}}dP$$

or, in more compact notation, $d(\Delta G_{\text{mix}}) = Td(\Delta S_{\text{mix}}) - d(\Delta H_{\text{mix}}) + d(\Delta PV)_{\text{mix}}$

$$d(\Delta G_{\text{mix}})_{T,P} = d(\Delta H_{\text{mix}})_{T,P} - Td(\Delta S_{\text{mix}})_{T,P} + d(\Delta PV)_{T,P}$$

or, what equation do we need if both ΔH_{mix} and ΔS_{mix} were known?

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Thermochemical Equations Study Guide Key

Christopher Talbot, Richard Harwood



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