

Chapter

7

THERMOCHEMISTRY

MULTIPLE CHOICE QUESTIONS

- Spontaneous reactions are: (a) Reversible (b) Irreversible (c) Non irreversible (d) None of these (LHR 05)
- The standard heat changes occur at: (a) 25°C and 2 atm (b) 298 K and 1 atm (c) 25°C and 1 mm hg (d) 273 K and 1 atm (BWP 08, GRW 06)
- $\text{H}^+ + \text{OH}^- \rightarrow \text{H}_2\text{O}$ the change in enthalpy for reaction is called: (a) Heat of reaction (b) heat of formation (c) Heat of neutralization (d) Heat of combustion (LHR 06)
- The net change in energy in a chemical reaction is same whether it takes place directly or indirectly. It is called: (a) Henry's law (b) Charlie's law (c) Hess's law (d) Graham's law (RWP 11, GRW 07, BWP 10)
- The enthalpy change when one mole of substance is completely burnt in excess of oxygen is called: (a) Enthalpy of atomization (b) Enthalpy of neutralization (c) Enthalpy of Combustion (d) Enthalpy of formation (LHR 07)
- Enthalpy change for the reaction: $\text{CH}_4(\text{g}) + 2\text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g}) + 2\text{H}_2\text{O}(\text{g})$ is called enthalpy of: (a) Formation (b) Combustion (c) Neutralization (d) Atomization (LHR 08)
- Standard enthalpies are measured at: (a) 273 K (b) 298K (c) 373 K (d) All of these (FSD 11, LHR 09)
- The exothermic process is: (a) Evaporation (b) Sublimation (c) Respiration (d) Boiling (DGK 10, GRW, LHR 11)
- A state function which describes together the internal energy and the product of pressure and volume is called: (a) Enthalpy (b) Internal energy (c) Work (d) Free energy (GRW 11)

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ACT



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Engineering Chemistry I (for BPUT) Patrizia Bocchetta, 2024 Internal Combustion Engines Shyam K. Agrawal, 2006 Salient Features The New Edition Is A Thoroughly Revised Version Of The Earlier Edition And Presents A Detailed Exposition Of The Basic Principles Of Design Operation And Characteristics Of Reciprocating I C Engines And Gas Turbines Chemistry Of Combustion Engine Cooling And Lubrication Requirements Liquid And Gaseous Fuels For Ic Engines Compressors Supercharging And Exhaust Emission Its Standards And Control Thoroughly Explained Jet And Rocket Propulsion Alternate Potential Engines Including Hybrid Electric And Fuel Cell Vehicles Are Discussed In Detail Chapter On Ignition System Includes Electronic Injection Systems For Si And Ci Engines 150 Worked Out Examples Illustrate The Basic Concepts And Self Explanatory Diagrams Are Provided Throughout The Text More Than 200 Multiple Choice Questions With Answers A Good Number Of Review Questions Numerical With Answers For Practice Will Help Users In Preparing For Different Competitive Examinations With These Features The Present Text Is Going To Be An Invaluable One For Undergraduate Mechanical Engineering Students And Amie Candidates *A Textbook of Physical Chemistry, 6th Edition* Sharma K.K. & Sharma L.K., 2016 A Textbook of Physical Chemistry **IIT Chemistry-II** , **AP Chemistry Premium, 2024: 6 Practice Tests + Comprehensive Review + Online Practice** Neil D. Jespersen, Pamela Kerrigan, 2023-07-04 Always study with the most up to date prep Look for AP Chemistry Premium 2025 Prep Book with 6 Practice Tests Comprehensive Review Online Practice ISBN 9781506291802 on sale July 2 2024 Publisher s Note Products purchased from third party sellers are not guaranteed by the publisher for quality authenticity or access to any online entities included with the product **Psychometrics** R. Michael Furr, 2017-11-17 Using a meaning based approach that emphasizes the why over the how to Psychometrics An Introduction provides thorough coverage of fundamental issues in psychological measurement Author R Michael Furr discusses traditional psychometric perspectives and issues including reliability validity dimensionality test bias and response bias as well as advanced procedures and perspectives including item response theory and generalizability theory The substantially updated Third Edition includes broader and more in depth coverage with new references a glossary summarizing over 200 key terms and expanded suggested readings consisting of highly relevant papers to enhance the book s overall accessibility scope and usability for both instructors and students Online Resources Free PowerPoint slides for instructors are available with this text Contact your rep to learn more Chemistry for Degree Students B.Sc. (Honours) Semester II, 1/e (As per CBCS) Madan R.L., 2022 This textbook has been designed to meet the needs of B Sc Honours Second Semester students of Chemistry as per the UGC Choice Based Credit System CBCS Maintaining the traditional approach to the subject this textbook lucidly explains the basics of Organic and Physical Chemistry Important topics such as alkanes alkenes alkynes stereochemistry aliphatic hydrocarbons thermochemistry chemical thermodynamics and chemical equilibrium are aptly discussed to give an overview of organic and physical

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