

Chemistry 30

Thermochemistry Review

1. Which of the following statements is **true** for an endothermic reaction?
- A. ΔH_r for the reaction is negative.
 - B. Energy is a product in the chemical equation.
 - C. The temperature of the surroundings increases.
 - D. The potential energy of the products is greater than the potential energy of the reactants.

Use the following information to answer the next question.



2. The ΔH_r value for the reverse reaction would be
- A. +100 kJ
 - B. +50.0 kJ
 - C. -50.0 kJ
 - D. -100 kJ

Numerical Response

1. Benzene is a gasoline additive. The heat of formation for 1.00 mol of $\text{C}_6\text{H}_6\text{(l)}$ is _____ kJ.

(Record your **three-digit answer** in the numerical-response section on the answer sheet)

Numerical Response

2. An outdoor mercury thermometer contains 1.02 g of mercury. On a winter day, the thermometer records a change in temperature from -38.0°C to -13.0°C . The mercury absorbed _____ J of energy. The heat capacity of mercury is $0.140 \text{ J/g}^\circ\text{C}$.

(Record your **three-digit answer** in the numerical-response section on the answer sheet)

Thermochemistry Review Sheet

Yi-Tong Ma



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