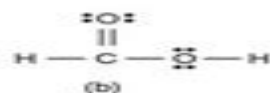
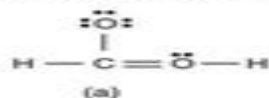


28. Carbon and oxygen can bond either by a single bond (as in $\text{CH}_3\text{-OH}$), a double bond (as in $\text{O}=\text{C}=\text{O}$), or a triple bond (as in $\text{HC}\equiv\text{CH}$).
- Describe these three types of bonds in terms of σ -bonds and π -bonds.
 - How would you expect the length of the carbon-oxygen bond to vary in the three given examples?
29. Explain the following in terms of the intermolecular forces that exist in them:
- Water is a liquid at room temperature and atmospheric pressure, but hydrogen sulphide is a gas.
 - At room temperature and atmospheric pressure chlorine is a gas, bromine is a liquid and iodine is a solid.
 - Pentan-1-ol boils at 137°C , whereas pentan-3-ol boils at 116°C .
 - The boiling point of sulphur dioxide is 24°C higher than that of chlorine.
30. Of the two possible valence-bond formulas for formic acid (HCO_2H), which one is the more reasonable structure?



31. Draw the correct valence-bond formula for hydrazine (N_2H_4).
32. Draw the resonance hybrid of SO_2 .
33. Based on the VB approach, predict the shape of a ClO_3^- ion.
34. The molecular orbital description of NO is similar to that of O_2 except that the energies of the oxygen atomic orbitals are slightly lower than those of nitrogen atomic orbitals and NO molecule contains different number electrons than O_2 .
- Draw an energy diagram to show how the atomic orbitals of N and O combine to form molecular orbitals of NO .
 - How many unpaired electrons are present in the NO molecule?
 - What is the bond order in NO molecule?
 - Is NO paramagnetic or diamagnetic?
35. List the following in order of increasing $\text{O}-\text{O}$ bond length : O_2 , O_2^+ , O_2^-
36. Show the direction of the bond moment (dipole moment) in each of the following bonds:



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1. Understanding the eBook Unit 3b Chemistry June 201
 - The Rise of Digital Reading Unit 3b Chemistry June 201
 - Advantages of eBooks Over Traditional Books
2. Identifying Unit 3b Chemistry June 201
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Unit 3b Chemistry June 201
 - User-Friendly Interface
4. Exploring eBook Recommendations from Unit 3b Chemistry June 201
 - Personalized Recommendations
 - Unit 3b Chemistry June 201 User Reviews and Ratings
 - Unit 3b Chemistry June 201 and Bestseller Lists
5. Accessing Unit 3b Chemistry June 201 Free and Paid eBooks

- Unit 3b Chemistry June 201 Public Domain eBooks
- Unit 3b Chemistry June 201 eBook Subscription Services
- Unit 3b Chemistry June 201 Budget-Friendly Options
- 6. Navigating Unit 3b Chemistry June 201 eBook Formats
 - ePub, PDF, MOBI, and More
 - Unit 3b Chemistry June 201 Compatibility with Devices
 - Unit 3b Chemistry June 201 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Unit 3b Chemistry June 201
 - Highlighting and Note-Taking Unit 3b Chemistry June 201
 - Interactive Elements Unit 3b Chemistry June 201
- 8. Staying Engaged with Unit 3b Chemistry June 201
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Unit 3b Chemistry June 201
- 9. Balancing eBooks and Physical Books Unit 3b Chemistry June 201
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Unit 3b Chemistry June 201
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Unit 3b Chemistry June 201
 - Setting Reading Goals Unit 3b Chemistry June 201
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Unit 3b Chemistry June 201
 - Fact-Checking eBook Content of Unit 3b Chemistry June 201
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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web im westen trennt der manhartsberg das weinviertel vom benachbarten waldviertel im süden bildet die donau die natürliche grenze die größte stadt des weinviertels ist stockerau daneben gibt es zahlreiche weitere städte und gemeinden in allen größenordnungen bezirkshauptstädte stadtgemeinden sowie gemeinden