

A light source of wavelength 520 nm emits 1.04×10^{15} photons per second while the second source of 460 nm produces 1.38×10^{15} photons per second. Then the ratio of power of second source to that of first source is

- a) 1.00
- b) 1.02
- c) 1.5
- d) 0.98

The mean wavelength of light from sun is taken to be 550 nm and its mean power is $3.8 \times 10^{26} \text{ W}$. The number of photons received by the human eye per second on the average from sunlight is of the order of

- a) 10^{15}
- b) 10^{12}
- c) 10^{24}
- d) 10^{21}

The threshold wavelength for a metal surface whose photoelectric work function is 3.313 eV is

- a) $4125 \text{ \AA}$
- b) $3750 \text{ \AA}$
- c) $6000 \text{ \AA}$
- d) $2062.5 \text{ \AA}$

A light of wavelength 500 nm is incident on a sensitive plate of photoelectric work function 1.235 eV . The kinetic energy of the photo electrons emitted is be (Take $h = 6.6 \times 10^{-34} \text{ Js}$)

- a) 0.58 eV
- b) 2.48 eV
- c) 1.24 eV
- d) 1.16 eV

Photons of wavelength λ are incident on a metal. The most energetic electrons ejected from the metal are bent into a circular arc of radius R by a perpendicular magnetic field having magnitude B . The work function of the metal is (KVPY-SX 2016)

- a. $\frac{hc}{\lambda} - m_e c^2 + \frac{e^2 B^2 R^2}{2m_e}$
- b. $\frac{hc}{\lambda} + 2m_e \left[\frac{eBR}{2m_e} \right]^2$

c. $\frac{hc}{\lambda} - m_e c^2 - \frac{e^2 B^2 R^2}{2m_e}$

d. $\frac{hc}{\lambda} - 2m_e \left[\frac{eBR}{2m_e} \right]^2$

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Table of Contents Unit Vii Physics

1. Understanding the eBook Unit Vii Physics
 - The Rise of Digital Reading Unit Vii Physics
 - Advantages of eBooks Over Traditional Books
2. Identifying Unit Vii Physics
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Unit Vii Physics
 - User-Friendly Interface
4. Exploring eBook Recommendations from Unit Vii Physics
 - Personalized Recommendations
 - Unit Vii Physics User Reviews and Ratings
 - Unit Vii Physics and Bestseller Lists
5. Accessing Unit Vii Physics Free and Paid eBooks
 - Unit Vii Physics Public Domain eBooks

- Unit Vii Physics eBook Subscription Services
- Unit Vii Physics Budget-Friendly Options
- 6. Navigating Unit Vii Physics eBook Formats
 - ePub, PDF, MOBI, and More
 - Unit Vii Physics Compatibility with Devices
 - Unit Vii Physics Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Unit Vii Physics
 - Highlighting and Note-Taking Unit Vii Physics
 - Interactive Elements Unit Vii Physics
- 8. Staying Engaged with Unit Vii Physics
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Unit Vii Physics
- 9. Balancing eBooks and Physical Books Unit Vii Physics
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Unit Vii Physics
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Unit Vii Physics
 - Setting Reading Goals Unit Vii Physics
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Unit Vii Physics
 - Fact-Checking eBook Content of Unit Vii Physics
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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