

10**CHEMICAL QUANTITIES****Chapter Test A****A. Matching**

Match each description in Column B to the correct term in Column A. Write the letter of the correct description on the line.

Column A

- _____ 1. standard temperature and pressure
- _____ 2. molar volume
- _____ 3. representative particle
- _____ 4. molar mass
- _____ 5. atomic mass
- _____ 6. Avogadro's number
- _____ 7. mole (mol)
- _____ 8. empirical formula
- _____ 9. percent composition
- _____ 10. molar mass

Column B

- a. the volume occupied by a mole of any gas at STP (22.4 L)
- b. the mass in amu of an atom of an element
- c. the number of representative particles in a mole of a substance
- d. the percent by mass of each element in a compound
- e. the mass of a mole of a substance
- f. the formula that gives the lowest whole-number ratio of the elements in a compound
- g. an atom, a formula unit, or a molecule
- h. the mass in grams of a mole of any pure substance
- i. 6.02×10^{23} representative particles of a substance
- j. 0°C and 101.3 kPa

B. Multiple Choice

Choose the best answer and write its letter on the line.

- _____ 11. How many hydrogen atoms are in 4 molecules of isopropyl alcohol, $\text{C}_3\text{H}_8\text{O}$?
 - a. $4 \times (6.02 \times 10^{23})$
 - b. 4
 - c. 32
 - d. $32 \times (6.02 \times 10^{23})$
- _____ 12. The mass of a mole of NaCl is its:
 - a. molar mass.
 - b. empirical formula.
 - c. gram atomic mass.
 - d. atomic mass.

Unit 6 Chapter 10 Chemical Quantities Test Review

Sabine Zange



Unit 6 Chapter 10 Chemical Quantities Test Review:

Unveiling the Power of Verbal Artistry: An Psychological Sojourn through **Unit 6 Chapter 10 Chemical Quantities Test Review**

In a global inundated with displays and the cacophony of fast communication, the profound energy and psychological resonance of verbal art usually fade into obscurity, eclipsed by the continuous assault of sound and distractions. Yet, located within the musical pages of **Unit 6 Chapter 10 Chemical Quantities Test Review**, a charming perform of literary splendor that pulses with fresh feelings, lies an remarkable trip waiting to be embarked upon. Written with a virtuoso wordsmith, this mesmerizing opus courses visitors on a psychological odyssey, delicately exposing the latent possible and profound influence stuck within the intricate internet of language. Within the heart-wrenching expanse of this evocative examination, we shall embark upon an introspective exploration of the book is key themes, dissect its captivating publishing design, and immerse ourselves in the indelible effect it leaves upon the depths of readers souls.

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