

## PRE TEST

## Chapter 6 Test-General/Honors Chemistry

## Part 1: Matching

Match each description in Column B with the correct term in Column A. Write the letter of the correct description on the line.

- |       |          |                           |                                                                                                              |
|-------|----------|---------------------------|--------------------------------------------------------------------------------------------------------------|
| C.2.2 | <u>a</u> | 1. cation                 | a. a positively charged ion                                                                                  |
| C.2.2 | <u>B</u> | 2. electronegativity      | b. an element whose highest occupied <i>s</i> or <i>p</i> sublevels are partially filled                     |
| C.2.5 | <u>J</u> | 3. inner transition metal | c. the energy required to remove an electron from an atom                                                    |
| C.2.6 | <u>I</u> | 4. alkali earth metal     | d. good conductor of heat and electric current                                                               |
| C.2.6 | <u>h</u> | 5. noble gas              | e. any element whose highest occupied <i>s</i> sublevel and a nearby <i>d</i> sublevel contain electrons     |
| C.2.5 | <u>b</u> | 6. representative element | f. any nonmetal in group 7A                                                                                  |
| C.2.2 | <u>d</u> | 7. metal                  | g. the tendency of an atom to attract electrons when it's in a compound (how much an atom wants an electron) |
| C.2.6 | <u>f</u> | 8. halogen                | h. an element in which the highest occupied <i>s</i> and <i>p</i> sublevels are completely filled            |
| C.2.5 | <u>e</u> | 9. transition metal       | i. an element in group 2A                                                                                    |
| C.2.2 | <u>c</u> | 10. ionization energy     | j. an element whose highest <i>s</i> sublevel and a nearby <i>f</i> sublevel contain electrons               |

## Part 2: Multiple Choice

Choose the best answer and write its letter on the line.

- |       |          |                                      |                          |                              |
|-------|----------|--------------------------------------|--------------------------|------------------------------|
| C.2.6 | <u>b</u> | 11. The alkali metals do not include | a. Li                    | c. Na                        |
|       |          |                                      | b. Ca                    | d. Rb                        |
| C.2.6 | <u>c</u> | 12. The element iodine, I, is a      | a. period 5 alkali metal | c. period 5 halogen          |
|       |          |                                      | b. period 4 halogen      | d. period 5 transition metal |

# Unit 6 Chapter Chemical Quantities Test Review

**D Siedentop**



## **Unit 6 Chapter Chemical Quantities Test Review:**

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