

Unit 3- thermochemistry

Study of Energy Changes that happen with matter changes

State Changes

Endothermic

S $\rightarrow$ g: Sublimation

S $\rightarrow$ l: melting

l $\rightarrow$ g: boiling

Exothermic

g $\rightarrow$ s: deposition

l $\rightarrow$ s: freezing

g $\rightarrow$ l: condensation

First law of thermodynamics
total amount of energy in the
universe is constant

System

↳ reactants + products

Surroundings

↳ all matter absorbing/releasing energy

System Types

open
mass + energy

closed
energy

isolated
neither

Type

Exchange



Universe

= System + Surroundings

Exothermic process

• energy is produced / released (reactants)

Endothermic Process

• energy is absorbed (product)

Unit 1test Thermochemistry

Darrell D. Ebbing



Unit 1test Thermochemistry:

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Scientific and Technical Aerospace Reports, 1994
Official Gazette Philippines, 2008

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