

SCIENCE 2 UNIT 3 CHEMISTRY

The Subjects :

- Bonding
 - Ionic Bonding
 - Covalent Bonding
- Nomenclature Sheet
 - Ionic Compounds
 - Multicharge metals
 - Molecular compounds
 - Polyatomic ions
- Physical vs Chemical Change
- Balancing
- Chemical reactions
- Gas Testing
- Acids & Base
- Acid Precipitation

Periodic table of the elements

Legend:

- Alkali metals
- Alkaline-earth metals
- Transition metals
- Other metals
- Nonmetals
- Noble gases
- Rare-earth elements (lanthanides and actinoids)
- Actinoid elements

Bonding

All atoms in a compound are held together with a chemical bond

Ionic Bonding

Attraction between electrons

Ionic compounds are made with a metal and non-metal and transferring electrons.

The metal loses electrons to become positive (cation).

Non-metal gains electrons to become negative (anion)

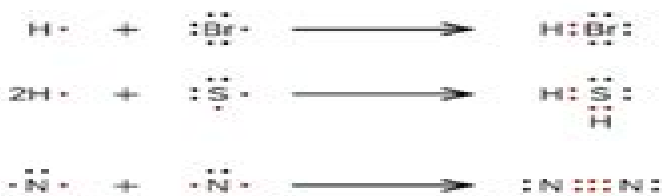


Covalent Bonding

Sharing between electrons

The result of a covalent bond is a molecule compound.

Molecule compounds are made with a non-metal and a non-metal.



Unit 7 3 Chemistry

CL Gary



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