Name: ______ Date: _____

MASS TO MOLE STOICHIOMETRY

Answer the questions. Write answers on new paper.

1. In photosynthesis, water reacts with carbon dioxide to give oxygen and glucose ($C_6H_{12}O_6$). Write and balance the chemical equation. How many moles of CO_2 are required to make 1200g of glucose?

a. What is the mass of potassium nitrate that is produced when 2.04 moles of potassium phosphate react?

b. If 5.80g of aluminum phosphate are formed, how many moles of aluminum nitrate reacted?

$$3. N_2 + 20_2 \longrightarrow N_2O_4$$

a If 150g of N_2O_4 was produced, how many moles of O_2 were required?

b. If 4.0×103 moles of oxygen reacted, how many grams of N_2 were needed?

a. If you have 5.50 mol of CaC2, how much C2H2 do you get?

b. How many moles of water are needed when 65.0g of CaC_2 have reacted?



Study Guide For The Mole Answers Chemistry

Iris Stovall

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