

10.1 Periodic Trends



PERIODIC TRENDS: FROM CONCEPT TO PRACTICE

Section 10.1 continued

In your notebook, read aloud the modern periodic table.

Copy the information in the box on the left below from the periodic table to complete the table on the right.

_____	19
_____	18
_____	Hydrogen
_____	Li (103)
_____	Neon (1018)

Atomic Mass	19.0
Atomic Number	18.0
Electron Configuration	1S ²
Chemical Name	Ne
Chemical Symbol	Ne

Put each item in Column A, either the letter of the matching item in Column B.

Column A

- _____ 19. It exists as the periodic table.
- _____ 20. It has the periodic table.
- _____ 21. Group 1 elements.
- _____ 22. Elements that are strong oxidizing agents.
- _____ 23. Group 18 elements.

Column B

- a. metals
- b. group
- c. period
- d. representative elements
- e. noble gases

In the space on the left, write each of the statements in terms of the statement in italics, change the underlined word as shown in *italic* form.

- _____ 24. There are two representative elements of elements.
- _____ 25. There are two periods of the elements in the periodic table.
- _____ 26. Group 1 elements except for hydrogen and lithium are the alkali metals.
- _____ 27. Group 18 elements are the noble gases.
- _____ 28. Group 1 elements are highly reactive elements in the periodic table.
- _____ 29. Group 18 elements are very reactive elements in the periodic table.
- _____ 30. Elements have properties of both metals and nonmetals.

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Robert R. Redfield



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