

Toxic Reactions: Chemical Equations

These activities are a part of the
Lab 5-4-2019 report



Purpose:

To interpret chemical equations involving toxic

Materials:

- Toxic Reactions table

Part II: Interpreting Chemical Equations

Chemical equation: $\text{H}_2\text{SO}_4 + \text{NaOH} \rightarrow \text{Na}_2\text{SO}_4 + \text{H}_2\text{O}$

Interpretation: A chemical equation represents a reaction between substances. The substances are represented by chemical formulas. The reaction is represented by an arrow pointing from the reactants to the products.

It is a chemical reaction that is represented by a chemical equation.

Symbol	Meaning in equation
H_2SO_4	Sulfuric acid
NaOH	Sodium hydroxide
$\rightarrow$	Reaction arrow
Na_2SO_4	Sodium sulfate
H_2O	Water

Symbol	Meaning in equation
H_2SO_4	Sulfuric acid
NaOH	Sodium hydroxide
$\rightarrow$	Reaction arrow
Na_2SO_4	Sodium sulfate
H_2O	Water

1. Write the chemical equation for the reaction between sulfuric acid and sodium hydroxide. $\text{H}_2\text{SO}_4 + \text{NaOH} \rightarrow \text{Na}_2\text{SO}_4 + \text{H}_2\text{O}$
2. Write the chemical equation for the reaction between sulfuric acid and sodium hydroxide. $\text{H}_2\text{SO}_4 + \text{NaOH} \rightarrow \text{Na}_2\text{SO}_4 + \text{H}_2\text{O}$

Interpreting Chemical Equations

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RC Schank



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