- On the basis of enthalpy and entropy, predict whether each of the following reactions would be spontaneous as written or not at room temperature.
  - 3 mole gas 2 moles gas a) No(8) + 200(8) # 2NO(8) AH = +67.7 kJ (ende)

Spontaneous as written? (yes-so) ...... Answer \_

- - Spontaneous as written? (yes/ac) ....... Answer yes, but it would
    - not go to completion. A state of equilibrium would be reached.

# **Unit 4 Section A Supplement Answers Chemistry**

Barbara C. Sorkin, Mahtab Jafari, Susan Murch, Connie M. Weaver

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