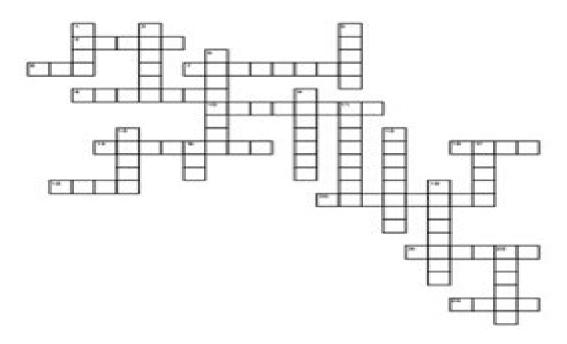
Atoms and Bonding



[4] An _____ is a mixture made of two or more elements, at least one of which is a metal.

[6] An ____ is an atom or group of atoms that has en electric charge.

[7] When a neutral alom loses a valence electron, it loses negative charge and becomes a ______ion.

[8] The ______ electrons of an atom are those electrons that have the highest energy.

[10] When a positive and negative ion bond, they form an ionic ______.

[14] The chemical bond formed when two aloms share electrons is called a ______ bond.

[16] lonic compounds have _____ melting points.

[18] Molecular compounds are _____ conductors.

[20] A _____ bond is the attraction between a positive metal ion and the electrons surrounding it.

[21] Metals conduct electric current easily because the valence electrons in a metal can move _____ among the atoms.

[23] A chemical _____ is the force of attraction that holds atoms together as a result of the arrangement of electrons between them.

Down

[1] Nonmetal atoms become stable when they ____ or share enough electrons to have eight valence electron
 [2] When two atoms share two pairs of electrons, it forms a ____ bond.
 [3] Group 13 elements have ___ valence electrons.
 [5] A ____ is a neutral group of atoms joined by covalent bonds.

[9] A chemical ____ is a group of symbols that shows the ratio of elements in a compound.

[11] When a neutral atom gains an electron, it gains negative charge and becomes a _____ ion.

[12] A covalent band in which electrons are shared unequally, is a _____ band.

[13] lonic solids are _____ that have an orderly, threedimensional arrangement.

[15] Molecular compounds have _____ melting points and boiling points.

[17] An _____ bond is the attraction between two oppositely charged ions.

[19] lonic compounds conduct _____ current when dissolved in water.

[22] Metal atoms react by _____ their valence electrons.

Study Of Bonds Between Atoms Crossword

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