

Chemical Bonding Review Worksheet

1. Which part of the atom is responsible for chemical bonding?

2. What are valence electrons (be specific)?

3. How many valence electrons do most atoms need to have a complete outer shell and be happy? _____
4. Which two elements only need two valence electrons to be happy?

5. Why do the elements you named in #4 only need two valence electrons?

6. How many valence electrons do elements in Group 1, the Alkali Metals, have?

7. How many valence electrons do elements in Group 2, the Alkaline Earth Metals, have?

8. What is an ion?

9. If an element gives away an electron, will it form a positive ion or a negative ion?

10. If an element gains an electron, will it form a positive ion or a negative ion?

11. What is the difference in a cation and an anion?

12. How do ionic bonds form?

13. How do covalent bonds form?

14. Bond the following atoms. Determine if they are ionic or covalent, circle your choice. Show the valence electrons and how they are either shared between the atoms or how they are transferred between atoms. Then write the chemical formula in the space provided.

Ionic or covalent		Ionic or covalent	
C	Cl	Mg	Cl
Formula _____		Formula _____	

Studyisland Chemical Structures And Bonding Answer Key

Arshad Iqbal



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Government Reports Announcements & Index, 1983-08 *Interference Problems on Wing-fuselage Combinations in Inviscid, Incompressible Flow* Aeronautical Research Council (Great Britain), J. Weber, M. Gaynor Joyce, 1974

Government Reports Annual Index, 1983 **Government Reports Annual Index: Keyword A-L**, 1983 *The Answer Key: A Comprehensive Explanation of Problem Solving Methods for General Chemistry Success (Volume One) (First Edition)* Rachel Turoscy, 2018-08-09

The Answer Key A Comprehensive Explanation of Problem Solving Methods for General Chemistry Success Volume 1 is a concise and accessible textbook that covers the critical information a student needs to understand the basic mathematics used in chemistry courses. The book provides easy to understand step by step instructions for solving general chemistry problems. The book begins with chapters dedicated to problem solving methodology and unit conversions. In subsequent chapters the text covers important topics like ionic and covalent bonding, chemical formula calculations, solubility and reactions in aqueous solution, gases, the first law of Thermodynamics, Quantum theory and electron configuration. It also covers periodic trends, the Lewis Dot Structures and bonding theories. Each chapter contains sample problems and practice problems to help further understanding of how math and chemistry go hand in hand. The Answer Key is an excellent resource for any undergraduate course that deals with the basic concepts of general chemistry.

[O Level Chemistry MCQ \(Multiple Choice Questions\)](#) Arshad Iqbal, 2019-06-27

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Chemistry 5.41 : Introduction to Structure, Bonding and Mechanism Daniel Schaeffer
Kemp, Massachusetts Institute of Technology, 1975

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