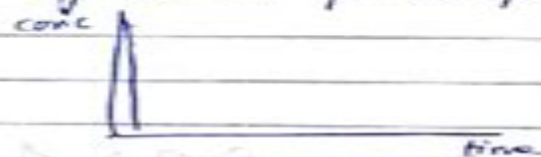


Chemical Kinetics

It mainly deals with analyses of conc. wrt time. (rate of reacⁿ), rate law, factors that can effect rate of reaction, mechanism of reaction.

On the basis of speed of reactions are divided in 3 types:

① Instantaneous reacⁿ: takes fraction of second for completion



② Eg Acid base neutralization reaction.
precipitation reaction.

② Very slow reacⁿ: Takes so many years for appreciable change in conc.
eg rusting of iron
eg conv. of graphite to diamond



③ Moderate reacⁿ: Takes ~~more~~ suitable time for appreciable change in conc.



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C Cleary



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