

Derive the following equations:

1. Show the following thermodynamic equations:

$$dU = TdS + PdV + \sum_i \mu_i dN_i \quad dH = TdS + VdP + \sum_i \mu_i dN_i$$

(a) is this a closed system? a closed system is a thermodynamic system

$$dU = TdS + PdV + \sum_i \mu_i dN_i \quad dH = TdS + VdP + \sum_i \mu_i dN_i$$

(b) what equation do we need to find the chemical potentials of the system? (Gibbs-Duhem equation)

$$dU = TdS + PdV + \sum_i \mu_i dN_i \quad dH = TdS + VdP + \sum_i \mu_i dN_i$$

(c) what equation do we need to find the chemical potentials of the system? (Gibbs-Duhem equation)

$$dU = TdS + PdV + \sum_i \mu_i dN_i \quad dH = TdS + VdP + \sum_i \mu_i dN_i$$

(d) what equation do we need to find the chemical potentials of the system? (Gibbs-Duhem equation)

$$dU = TdS + PdV + \sum_i \mu_i dN_i \quad dH = TdS + VdP + \sum_i \mu_i dN_i$$

2. Derive the following thermodynamic equations: $dU = TdS + PdV + \sum_i \mu_i dN_i$ and $dH = TdS + VdP + \sum_i \mu_i dN_i$

is this a closed system? a closed system is a thermodynamic system which does not exchange matter with the surroundings. The thermodynamic system is a closed system.

$$dU = TdS + PdV + \sum_i \mu_i dN_i \quad dH = TdS + VdP + \sum_i \mu_i dN_i$$

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$$dU = TdS + PdV + \sum_i \mu_i dN_i \quad dH = TdS + VdP + \sum_i \mu_i dN_i$$

3. Derive the following thermodynamic equations: $dU = TdS + PdV + \sum_i \mu_i dN_i$ and $dH = TdS + VdP + \sum_i \mu_i dN_i$

a. Show a thermodynamic equation for the system.

$$dU = TdS + PdV + \sum_i \mu_i dN_i \quad dH = TdS + VdP + \sum_i \mu_i dN_i$$

is this a closed system? a closed system is a thermodynamic system which does not exchange matter with the surroundings. The thermodynamic system is a closed system.

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