Thermochemistry Practice Problems



Flow of Energy:

1. How many kJ of energy are in a donut that contains 200.0 Calories?

 What is the specific heat of a substance that has a mass of 25.0 g and requires 525.0 calories to raise its temperature by 15.0°C?

$$m = 25.0_{9}$$

 $q = 525.0 \text{ cal}$ $525.0 \text{ cal} = (25.0 \text{ g}) \times (25.0 \text{ c})$
 $\Delta T = 15.0 \text{ c}$ $C = 1.40^{\text{cal}}/\text{g} \cdot \text{c}$

Suppose 100.0 g of H₂O(s) absorbs 1255.0 J of heat. What is the corresponding temperature change?
 The specific heat capacity of H₂O(s) is 2.1 J/g°C. Q = mC Δ T

 How many joules of heat energy are required to raise the temperature of 100.0 g of aluminum by 120.0°C? The specific heat capacity of aluminum is 0.90 J/g°C. Q = MC A T

5. In reference to #4, what is the molar heat capacity of aluminum?

Measuring and Expressing Heat Changes:

6. A student mixed 75.0 mL of water containing 0.75 mol HCl at 25°C with 75.0 mL of water containing 0.75 mol of NaOH at 25°C in a foam cup calorimeter. The temperature of the resulting solution increased to 35°C. How much heat in kilojoules was released by this reaction (Course = 4.18 J/g°C)?

Calculate the amount of heat evolved when 150 g of water are reacted with excess calcium according to the following thermochemical equation:

$$Ca(s) + 2H_2O(l) \rightarrow Ca(OH)_2(s) + H_2(g)$$
 $\Delta H = -65.2 \text{ kJ}$
 $150_3 \text{ HeO} \mid Imol \text{ HeO} \mid -65.2 \text{ kJ} = \left[-270 \text{ kJ} \right] \text{ exothermic}$

Thermochemistry Practice Problems 2 Answers

Arshad Iqbal

Thermochemistry Practice Problems 2 Answers:

Survival Guide to General Chemistry Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning acid base reactions which often persists through working with acid base equilibrium Many chapters provide alternative viewpoints as an aid to understanding This book addresses a very real need for a large number of incoming freshman in STEM fields Chemistry Workbook For Dummies Peter J. Mikulecky, Katherine Brutlag, Michelle Rose Gilman, Brian Peterson, 2008-08-06 From liquids and solids to acids and bases work chemistry equations and use formulas with ease Got a grasp on the chemistry terms and concepts you need to know but get lost halfway through a problem or worse yet not know where to begin Have no fear this hands on guide helps you solve many types of chemistry problems in a focused step by step manner With problem solving shortcuts and lots of practice exercises you ll build your chemistry skills and improve your performance both in and out of the science lab You ll see how to work with numbers atoms and elements make and remake compounds understand changes in terms of energy make sense of organic chemistry and more 100s of Problems Know where to begin and how to solve the most common chemistry problems Step by step answer sets clearly identify where you went wrong or right with a problem Understand the key exceptions to chemistry rules Use chemistry in practical applications with confidence **Student Solutions Manual** Darrell Ebbing, 2005-03-17 Provides solutions to odd numbered Practice Problems General Problems and Cumulative Skills Problems plus answers to Review Ouestions Chemistry: 1001 Practice Problems For Dummies (+ Free Online Practice) Heather Hattori, Richard H. Langley, 2022-05-10 Practice your way to a better grade in your Chemistry class Chemistry 1001 Practice Problems For Dummies gives you 1 001 opportunities to practice solving problems on all the topics covered in your chemistry class in the book and online Get extra practice with tricky subjects solidify what you ve already learned and get in depth walk throughs for every problem with this useful book These practice problems and detailed answer explanations will catalyze the reactions in your brain no matter what your skill level Thanks to Dummies you have a resource to help you put key concepts into

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