

Thermochemistry Problems - Worksheet Number One (answers available on web site)

1. How much energy must be absorbed by 20.0 g of water to increase its temperature from 283.0 °C to 303.0 °C?
2. When 15.0 g of steam drops in temperature from 275.0 °C to 250.0 °C, how much heat energy is released?
3. How much energy is required to heat 120.0 g of water from 2.0 °C to 24.0 °C?
4. If 720.0 g of steam at 400.0 °C absorbs 800.0 kJ of heat energy, what will be its increase in temperature?
5. How much heat (in kJ) is given out when 85.0 g of lead cools from 200.0 °C to 10.0 °C? (C_p of lead = 0.129 J/g °C)
6. If it takes 41.72 joules to heat a piece of gold weighing 18.69 g from 10.0 °C to 27.0 °C, what is the specific heat of the gold?
7. It takes 333.51 joules to melt exactly 1 gram of H₂O. What is the molar heat of fusion for water, from this data?
8. A certain mass of water was heated with 41,840 Joules, raising its temperature from 22.0 °C to 28.5 °C. Find the mass of water.
9. How many joules of heat are needed to change 50.0 grams of ice at -15.0 °C to steam at 120.0 °C. Make a graph to indicate this change.
10. Calculate the number of joules given off when 32.0 grams of steam cools from 110.0 °C to ice at -40.0 °C. Make a graph to indicate this change.
11. You have a sample of H₂O with a mass 23.0 grams at a temperature of -46.0 °C. How many kilojoules of heat energy are necessary to carry out each step? Also, please calculate the total amount of energy needed and make a time-temperature graph.
 - a) heat the ice to 0.0 °C?
 - b) melt the ice?
 - c) heat the water from 0.0 °C to 100.0 °C?
 - d) boil the water?
 - e) heat the steam from 100.0 °C to 109.0 °C?
12. Calculate the energy released when 10.0 g of steam at 120.0 °C are converted into ice at minus 20.0 °C.
13. How much energy is required to convert 100.0 g of water at 20.0 °C completely to steam at 100.0 °C?
14. What amount of ice must be added to 540.0 g of water at 25.0 °C to cool the water to 0.0 °C and have no ice remaining?
15. How many grams of ice could be melted by the energy obtained as 18.0 g of steam is condensed at 100.0 °C and cooled to 0.0 °C?
16. If 150.0 grams of iron at 95.0 °C, is placed in an insulated container containing 500.0 grams of water at 25.0 °C, and both are allowed to come to the same temperature, what will that temperature be? The specific heat of water is 4.18 J/g °C and the specific heat of iron is 0.444 J/g °C)
17. When 80.0 grams of a certain metal at 90.0 °C was mixed with 100.0 grams of water at 30.0 °C, the final equilibrium temperature of the mixture was 36.0 °C. What is the specific heat (Joules gram⁻¹ °C⁻¹) of the metal?
18. Calculate the specific heat of a metal if a 55.0 g sample of an unknown metal at 99.0 °C causes a 1.7 °C temperature rise when added to 225.0 g of water at 22.0 °C.
19. If 10.0 g water at 0.0 °C is mixed with 20.0 g of water at 30.0 °C, what is the final temperature of a mixture?
20. 15.0 g of water at 0.0 °C are added to 40.0 g of water at 40.0 °C. What is the final temperature of the mixture?

Thermochemistry Problems Worksheet Number 1

**Bull, Prince Hycy, Patterson, Gerrelyn
Chunn**

A red circular graphic with a gradient, appearing as a partial circle or a thick arc, located to the right of the author names.

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General Chemistry Workbook Daniel C. Tofan,2010-07-28 This workbook is a comprehensive collection of solved exercises and problems typical to AP introductory and general chemistry courses as well as blank worksheets containing further practice problems and questions It contains a total of 197 learning objectives grouped in 28 lessons and covering the vast majority of the types of problems that a student will encounter in a typical one year chemistry course It also contains a fully solved 50 question practice test which gives students a good idea of what they might expect on an actual final exam covering the entire material

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Learning about Materials Ted Lister,Colin Osborne,1998 This book presents useful material in the form of teacher's notes and photocopiable worksheets It will be useful at pre and post 16 level

Proceedings of the International Conference on Social Science and Education (ICoESSE 2023) Jhoni Warmansyah,Artha Nesa Chandra,M. Haviz,Isra Nurmai Yenti,Safrizal Safrizal,Hadiyati Idrus,2023-11-16 This is an open access book We would like to invite you to The 1st International Conference Social Science and Education ICoESSE 2023 will be conducted on September 6-7 2023 at the Universitas Islam Negeri Mahmud Yunus Batusangkar West Sumatera Indonesia The aim of the Conference is to bring together scholars students researcher and administrators from different countries and to discuss theoretical and practical

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41st AIAA Aerospace Sciences Meeting & Exhibit ,2003

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Bridging School and University M. Cantrell,1993

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Experimental Thermochemistry. Vol.1. Measurement of Heats of Reaction F. D. Rossini, International Union of Pure and Applied Chemistry,

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