

Chemistry 30

Thermochemistry Quiz #2 Practice

Use the following information to answer the next question:

The combustion of ethanol ($\text{C}_2\text{H}_5\text{OH}$) and acetone ($\text{C}_3\text{H}_8\text{O}$) can be represented by the following thermochemical equations:



Equation 1



Equation 2



Equation 3

4. The enthalpy change for the combustion of ethanol as shown in equation 1 is

- a. -1366.8 kJ
- b. -2021.2 kJ
- c. -40.1 kJ
- d. -1426.7 kJ

Use the following information to answer the next question:

When ethanol, $\text{C}_2\text{H}_5\text{OH}$, combusts, the following equation



4. The molar enthalpy of formation, ΔH_f° , for ethanol is

- a. -1366.8 kJ/mol
- b. -2021.2 kJ/mol
- c. -40.1 kJ/mol
- d. -1426.7 kJ/mol

Thermochemistry Question Answer Guide

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Thermochemistry Question Answer Guide:

Unveiling the Magic of Words: A Review of "**Thermochemistry Question Answer Guide**"

In a world defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Thermochemistry Question Answer Guide**," a mesmerizing literary masterpiece penned with a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

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