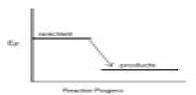
- 7. A chemical system that has no output or input of matter but does have an input/output of energy is known as:
  - a. A closed system
  - b. An open system

- c. An isolated system
- d. None of the above

- 8. Which statement concerning the diagram below is true?
  - a. AH is positive
  - b. The system is endothermic
  - c. Energy is released to the surroundings
  - d. The enthalpy of the products is greater than the enthalpy of the reactants



- In the reaction 2 NO (g) + O<sub>2</sub> (g) -> 2 NO<sub>2</sub> (g), if the concentration of nitrogen dioxide changes from 0.45 mol/L to 1.00 mol/L in 2 minutes, what is the overall rate of production of nitrogen dioxide in the system?
  - a. 3.64 mol/(L•min)
  - b. 0.275 mol/(L+min)
  - c. 0.137 mol/(L•min)

- d. 12.6 mol/(L•min)
- e. 0.333 mol/(L-min)

- 10. The activated complex...
  - a. is an unstable molecule
  - b. has the maximum potential energy possible
  - c. may revert to reactants
  - d. all of the above
- 11. A catalyst works by:
  - a. changing the products that are formed in the reaction
  - b. decreasing the activation energy of the reaction
  - c. decreasing the enthalpy change of the reaction
  - d. changing the enthalpy of the reactants
  - e. increasing the activation energy of the reaction
- 12. Which of the following could be a rate law for a fourth order reaction?
  - a.  $r = k[X]^4[Y]^1$
  - b.  $r = k[X]^4[Y]^4$

- c.  $r = k[X]^2[Y]^2$
- d.  $r = k[X]^3[Y]^2$
- 13. What property could help indicate the reaction rate for the following reaction?
  - $NaOH_{(s)} \rightarrow Na^{+}_{(mq)} + OH^{-}_{(mq)}$

ΔH = -44.51 kJ/mol

- a. Change in pressure
- b. Change in color
- c. Change in temperature
- d. Change in mass

# **Thermodynamics And Reaction Rates Review Answer**

**Robert R. Redfield** 

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