

Thermochemistry Practice Problems

- 10) How much work in joules is done when a piston expands from a volume of 13.27 liters to 76.55 liters against a pressure of 14.89 atm?

10) _____

$$-15450 \text{ J}$$

- 11) How much work in joules is done on the system when a 1.15 atm external pressure causes a piston to decrease in volume from 6.55 liters to 3.16 liters?

11) _____

$$395 \text{ J}$$

- 12) In a refrigeration system, the refrigerant gas absorbs 21.39 kJ of energy while expanding against a 0.278 atmosphere pressure from a volume of 0.0423 liters to a volume of 1.876 liters. What is the energy change of the gas?

12) _____

$$21.338 \text{ kJ}$$

- 13) What is the energy change of the system if a chemical reaction transfers 32.146 kJ of heat to the surroundings while it causes the expansion of a 1.465 liter vessel to 3.687 liters against a pressure of 3.64 atmospheres?

13) _____

$$\Delta E = -32.965 \text{ kJ}$$

- 14) Lead, water, sulfur, and arsenic have specific heats of 0.128, 4.18, 0.706, and 0.329 J/g°C, respectively. Which of these would require the smallest amount of heat to increase its temperature by 10 °C (assume all samples have the same mass)?

14) _____

LEAD $\Rightarrow$ SMALLEST SPECIFIC HEAT

- 15) When power was turned off to a 30.0 gal. water heater, the temperature of the water dropped from 75.0 °C to 22.5 °C. How much heat was lost to the surroundings? (1 gal = 3.785 L)

15) _____

$$2.49 \times 10^4 \text{ kJ}$$

Thermochemistry Problem Set One Specific Heat

Kaplan Test Prep

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular shape partially visible behind it.

Thermochemistry Problem Set One Specific Heat:

Enjoying the Song of Appearance: An Mental Symphony within **Thermochemistry Problem Set One Specific Heat**

In a global taken by screens and the ceaseless chatter of quick interaction, the melodic beauty and psychological symphony produced by the written word usually disappear in to the back ground, eclipsed by the persistent noise and disruptions that permeate our lives. But, situated within the pages of **Thermochemistry Problem Set One Specific Heat** a charming fictional treasure full of organic thoughts, lies an immersive symphony waiting to be embraced. Crafted by a masterful musician of language, this interesting masterpiece conducts visitors on an emotional journey, skillfully unraveling the hidden songs and profound influence resonating within each cautiously constructed phrase. Within the depths of the poignant review, we can discover the book is central harmonies, analyze its enthralling writing design, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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