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Zhou Lu section 313

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Thin Layer Chromatography

Objective: This experiment uses thin-layer chromatography to determine the solvent polarity effects and to identify an unknown substance. Retention factors will be calculated to determine the identity of the unknown compounds.

Materials: Breakers, flasks, stir rod, mini capillaries tubes, graduated cylinder, TLC plates, filter paper, Aluminum foil top, watch glass, tweezers, UV light, developing chamber, ruler, pencil, way paper, metal spatula, Acetaminophen, Aspirin, Caffeine, Ibuprofen (TBD), Ethanol, Dichloromethane, Acetic Acid, Ethyl Acetate, Hexanes.

Table:

Sample name	Chemical Formula	Molar Mass	Density	Melting Point	Solubility in Ethanol
Ethanol	C ₂ H ₅ OH	46.07 g/mol	789 kg/m ³	-144.1 °C	soluble
Acetaminophen	C ₈ H ₉ NO ₂	151.163 g/mol	1.26 g/cm ³	169 °C	Slightly soluble
Caffeine	C ₈ H ₁₀ N ₄ O ₂	194.19 g/mol	1.23 g/cm ³	227-228 °C	Slightly soluble
Ibuprofen	C ₁₃ H ₁₈ O ₂	206.29 g/mol	1.03 g/cm ³	75-78 °C	Soluble
Aspirin	C ₉ H ₈ O ₄	180.158 g/mol	1.4 g/cm ³	135 °C	Slightly Soluble
Dichloromethane	CH ₂ Cl ₂	84.93 g/mol	1.33 g/cm ³	-96.7 °C	Slightly Soluble

Thin Layer Chromatography Lab Report

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