Name: Jaiden Santo Date: March. 16th

Student Exploration: Calorimetry Lab

Vocabulary: calorie, calorimeter, joule, specific heat capacity

Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

- The Latin word calor means "heat," and meter comes from the Greek word meaning "to measure." What do you think a calorimeter does? The calorimeter is an instrument used to measure heat or heat energy and any changes in it.
- 2. Where have you heard the word calorie before? What do you think a calorie is? When I hear calorie I think it's about losing weight with your foods. I think a calorie is an ingredient, or type of energy. I also know that people sometimes tend to worry about how much calories they intake, and whatnot. And I also know that you can burn calories, which can suggest that it is a type of energy.

Gizmo Warm-up

A calorimeter is an insulated container filled with a liquid, usually water. When a hot object is placed in the calorimeter, heat energy is transferred from the object to the water and the water heats up.

Calorimeters can be used to find a substance's **specific heat capacity**. You will use the Calorimetry Lab Gizmo to determine the specific heat capacities of various substances.



- On the SIMULATION pane, select Copper. Use the slider to set its Mass to 200 g. Set the Water mass to 200 g. Check that the Water temp is set to 30.0 °C and the copper's Temp is 90 °C. Select the GRAPH tab, and click Play (**).
 - A. What was the Final temperature of the copper and the water? 34,96°C
 - B. How much did the temperature of the copper change? 55.04

C. How much did the temperature of the water change? 4.96
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Table of Contents Student Exploration Calorimetry Lab Answer Key

- 1. Understanding the eBook Student Exploration Calorimetry Lab Answer Key
 - The Rise of Digital Reading Student Exploration Calorimetry Lab Answer Key
 - Advantages of eBooks Over Traditional Books
- 2. Identifying Student Exploration Calorimetry Lab Answer Key
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
- 3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Student Exploration Calorimetry Lab Answer Key
 - User-Friendly Interface
- 4. Exploring eBook Recommendations from Student Exploration Calorimetry Lab Answer Key
 - Personalized Recommendations

- Student Exploration Calorimetry Lab Answer Key User Reviews and Ratings
- Student Exploration Calorimetry Lab Answer Key and Bestseller Lists
- 5. Accessing Student Exploration Calorimetry Lab Answer Key Free and Paid eBooks
 - Student Exploration Calorimetry Lab Answer Key Public Domain eBooks
 - Student Exploration Calorimetry Lab Answer Key eBook Subscription Services
 - Student Exploration Calorimetry Lab Answer Key Budget-Friendly Options
- 6. Navigating Student Exploration Calorimetry Lab Answer Key eBook Formats
 - ∘ ePub, PDF, MOBI, and More
 - Student Exploration Calorimetry Lab Answer Key Compatibility with Devices
 - Student Exploration Calorimetry Lab Answer Key Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Student Exploration Calorimetry Lab Answer Key
 - Highlighting and Note-Taking Student Exploration Calorimetry Lab Answer Key
 - Interactive Elements Student Exploration Calorimetry Lab Answer Key
- 8. Staying Engaged with Student Exploration Calorimetry Lab Answer Key
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Student Exploration Calorimetry Lab Answer Key
- 9. Balancing eBooks and Physical Books Student Exploration Calorimetry Lab Answer Key
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Student Exploration Calorimetry Lab Answer Key
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Student Exploration Calorimetry Lab Answer Key
 - Setting Reading Goals Student Exploration Calorimetry Lab Answer Key
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Student Exploration Calorimetry Lab Answer Key
 - Fact-Checking eBook Content of Student Exploration Calorimetry Lab Answer Key

- Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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