# 11 BALANCING CHEMICAL EQUATIONS

Directions:. Label decomposition, synthesis (combination), single displacement, double displacement (or double replacement), combustion reaction or Neutralization (acid/base).

Balance the Chemical Equation	Reaction Type
1. 2 K + 1 Cl <sub>2</sub> → 2 KCl	COMBINATION
2. $\_SO_2 + \_O_2 \rightarrow \_SO_3$	S
3KCIO <sub>3</sub> →KCI +O <sub>2</sub>	<u> </u>
4CO <sub>2</sub> +C →CO	e <del></del>
5H <sub>2</sub> +O <sub>2</sub> →H <sub>2</sub> O	= <del>-</del>
6MgCO <sub>3</sub> →MgO +O <sub>2</sub>	-
7. $\underline{\hspace{0.1cm}}$ $H_2O + \underline{\hspace{0.1cm}}$ $Cs \rightarrow \underline{\hspace{0.1cm}}$ $CsOH + \underline{\hspace{0.1cm}}$ $H_2$	<u> </u>
8. $_{\text{Fe}_2\text{O}_3} + _{\text{Al}} \rightarrow _{\text{Fe}} + _{\text{Al}_2\text{O}_3}$	S <u>.                                    </u>
9PbCl₂ + _Li₂SO₄ → _PbSO₄ + _LiCl	82
10. $\_NaOH + \H_2SO_4 \rightarrow \Na_2SO_4 + \H-OH$	
12Pb(OH) <sub>2</sub> +HCl →PbCl <sub>2</sub> +H-OH	<u></u>
13Na <sub>2</sub> CO <sub>3</sub> +HCl →H <sub>2</sub> CO <sub>3</sub> +NaCl	<u> </u>
14AgNO <sub>3</sub> +Cu →Ag +Cu(NO <sub>3</sub> ) <sub>2</sub>	-
15. $\_CH_4 + \_O_2 \rightarrow \_CO_2 + \_H_2O$	S:
16. $C_2H_6 + O_2 \rightarrow CO_2 + H_2O$	-
17. $\_C_3H_8 + \_O_2 \rightarrow \_CO_2 + \_H_2O$	
18. $\_C_4H_{10} + \_O_2 \rightarrow \_CO_2 + \_H_2O$	
19. $\_C_5H_{12} + \_O_2 \rightarrow \_CO_2 + \_H_2O$	·

Bonus: Don't unless given permission. Use decimal method if needed, for example "1.5 O2."

20. 
$$C_7H_5N_3O_6 + O_2 \rightarrow CO_2 + H_2O + N_2$$

# **Synthesis Reaction Worksheet With Answers**

**M Woodhall** 

**Synthesis Reaction Worksheet With Answers:** 

This book delves into Synthesis Reaction Worksheet With Answers. Synthesis Reaction Worksheet With Answers is a vital topic that needs to be grasped by everyone, ranging from students and scholars to the general public. The book will furnish comprehensive and in-depth insights into Synthesis Reaction Worksheet With Answers, encompassing both the fundamentals and more intricate discussions.

- 1. This book is structured into several chapters, namely:
  - Chapter 1: Introduction to Synthesis Reaction Worksheet With Answers
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  - o Chapter 3: Synthesis Reaction Worksheet With Answers in Everyday Life
  - Chapter 4: Synthesis Reaction Worksheet With Answers in Specific Contexts
  - ∘ Chapter 5: Conclusion
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- 3. In chapter 2, the author will delve into the foundational concepts of Synthesis Reaction Worksheet With Answers. This chapter will elucidate the essential principles that must be understood to grasp Synthesis Reaction Worksheet With Answers in its entirety.
- 4. In chapter 3, this book will examine the practical applications of Synthesis Reaction Worksheet With Answers in daily life. The third chapter will showcase real-world examples of how Synthesis Reaction Worksheet With Answers can be effectively utilized in everyday scenarios.
- 5. In chapter 4, this book will scrutinize the relevance of Synthesis Reaction Worksheet With Answers in specific contexts. This chapter will explore how Synthesis Reaction Worksheet With Answers is applied in specialized fields, such as education, business, and technology.
- 6. In chapter 5, this book will draw a conclusion about Synthesis Reaction Worksheet With Answers. The final chapter will summarize the key points that have been discussed throughout the book.

  This book is crafted in an easy-to-understand language and is complemented by engaging illustrations. It is highly

recommended for anyone seeking to gain a comprehensive understanding of Synthesis Reaction Worksheet With Answers.

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