

Unit 3 Cellular Energetics AP Biology Exam Review

Enzyme Structure

- 1) To the right is an enzyme-substrate complex. Label what A, B, and C represent.
- 2) Describe how enzymes work. Make sure to include the terms products, reactants, and activation energy.

The activation energy is the energy required to start a reaction. Enzymes are proteins that bind to a molecule, or substrate, to modify it and lower the energy required to make it react. Reactants (or 'substrates') are the starting materials for a reaction. ... Because they speed up reactions, they are called catalysts.



Enzyme Catalysis and Environmental Effects

- 3) How can a scientist make an enzyme reaction increase in production rates?

Enzyme concentration: Increasing enzyme concentration will speed up the reaction, as long as there is substrate available to bind to. ... Substrate concentration: Increasing substrate concentration also increases the rate of reaction to a certain point.

- 4) What could slow down the creation of products from an enzymatic reaction? Give at least three items.

Temperature: Raising temperature generally speeds up a reaction, and lowering temperature slows down a reaction.

- 5) What is denaturation? What causes this process to occur?

If a protein loses its shape, it ceases to perform that function. The process that causes a protein to lose its shape is known as denaturation. Denaturation is usually caused by external stress on the protein, such as solvents, inorganic salts, exposure to acids or bases, and by heat.

pH Math

- 4) What's the concentration of a HCl solution that has a pH of 3.5?

Cell Energy: First and Second Law of Thermodynamics

- 5) Compare the first law to the second law of thermodynamics. Give examples of each.

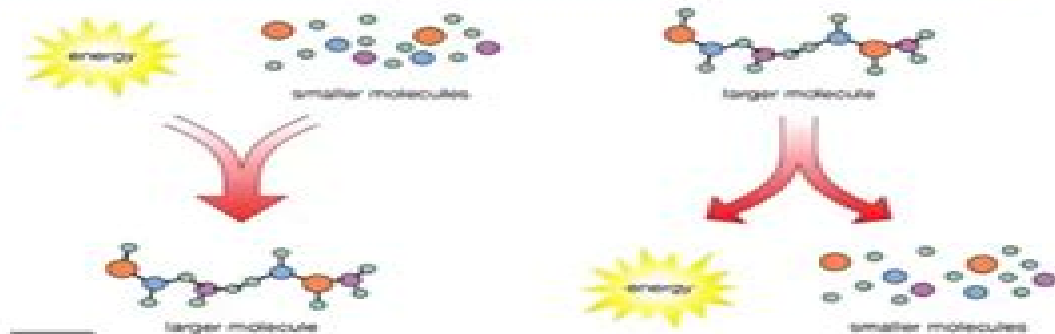
The first law, also known as Law of Conservation of Energy, states that energy cannot be created or destroyed in an isolated system. The second law of thermodynamics states that the entropy of any isolated system always increases.

Cell Energy: Catabolic and Anabolic Reactions

- 6) Compare catabolic to anabolic reactions.

Anabolism refers to the process which builds molecules the body needs; it usually requires energy for completion. Catabolism refers to the process that breaks down complex molecules into smaller molecules; it usually releases energy for the organism to use.

- 7) Look at the picture to the right. Which diagram (A or B) is an anabolic reaction and which one is a catabolic reaction. How can you tell?



$$\text{pH} = -\log[\text{H}^+]$$
$$[\text{H}^+] = 10^{-\text{pH}}$$

Unit 3gas Heat Review Answers

Minjie Lin

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Unit 3gas Heat Review Answers

1. Understanding the eBook Unit 3gas Heat Review Answers
 - The Rise of Digital Reading Unit 3gas Heat Review Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Unit 3gas Heat Review Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in a Unit 3gas Heat Review Answers
 - User-Friendly Interface
4. Exploring eBook Recommendations from Unit 3gas Heat Review Answers
 - Personalized Recommendations
 - Unit 3gas Heat Review Answers User Reviews and Ratings

- Unit 3gas Heat Review Answers and Bestseller Lists
- 5. Accessing Unit 3gas Heat Review Answers Free and Paid eBooks
 - Unit 3gas Heat Review Answers Public Domain eBooks
 - Unit 3gas Heat Review Answers eBook Subscription Services
 - Unit 3gas Heat Review Answers Budget-Friendly Options
- 6. Navigating Unit 3gas Heat Review Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Unit 3gas Heat Review Answers Compatibility with Devices
 - Unit 3gas Heat Review Answers Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Unit 3gas Heat Review Answers
 - Highlighting and Note-Taking Unit 3gas Heat Review Answers
 - Interactive Elements Unit 3gas Heat Review Answers
- 8. Staying Engaged with Unit 3gas Heat Review Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Unit 3gas Heat Review Answers
- 9. Balancing eBooks and Physical Books Unit 3gas Heat Review Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Unit 3gas Heat Review Answers
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Unit 3gas Heat Review Answers
 - Setting Reading Goals Unit 3gas Heat Review Answers
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Unit 3gas Heat Review Answers
 - Fact-Checking eBook Content of Unit 3gas Heat Review Answers
 - Distinguishing Credible Sources

13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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