

Name: _____

Nuclear Chemistry (Fusion, Fission, Half-life, Radioactive Decay) - Worksheet

Q.1. What is the name of the process of emitting energy and subatomic particles spontaneously by unstable nuclei?

- a) Metastability
- b) Conductivity
- c) Half-life
- d) Radioactivity

Q.2. Most nuclear fission reactions are initiated by bombarding nuclei with _____

- a) electrons
- b) protons
- c) neutrons
- d) positrons

Q.3. Identify the following as nuclear fusion or nuclear fission.

	
• The combining of two or more small nuclei to form a larger nucleus.	• The splitting of a large nucleus into smaller pieces.

Q.4. Which kind of nuclear reaction is used in power plants to generate electrical energy?

- a) Nuclear Fusion
- b) Nuclear Fission
- c) Combination reaction
- d) Decomposition reaction

Q.5. Which process produces energy in stars, such as our Sun?

- a) Nuclear Fusion
- b) Nuclear Fission
- c) Combination reaction
- d) Decomposition reaction

Q.6. Write T or True if the statement is true, write F or False if the statement is false.

- _____ 1. Nuclear reactions don't affect the nucleus of an atom.
- _____ 2. The rate of radioactive decay of a specific radioactive isotope depends on external factor s such as temperature and pressure.
- _____ 3. Nuclear fusion generally releases more energy than nuclear fission.
- _____ 4. In nuclear reactions, atomic number and mass number are balanced instead of atoms.

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J Ma



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