

Chapter 20
Electrons in Atoms
Worksheet

Name _____ Class _____

State _____

Write the quantum numbers defining each of the seven orbitals

Write quantum numbers for, describe, and give magnetic moment and orbital angular momentum for, s, p, d, and f orbitals, and the quantum numbers for the orbitals in the hydrogen atom.

Questions

Answers

1. Is the electron in the 1s orbital the same as the electron in the 2s orbital?
No, it is not.

Question

2. Is the electron in the 1s orbital the same as the electron in the 2s orbital?
No, it is not.

Answer

3. Is the electron in the 1s orbital the same as the electron in the 2s orbital?
No, it is not.

Answer

4. Is the electron in the 1s orbital the same as the electron in the 2s orbital?
No, it is not.

Answer

5. Is the electron in the 1s orbital the same as the electron in the 2s orbital?
No, it is not.

Answer

6. Is the electron in the 1s orbital the same as the electron in the 2s orbital?
No, it is not.

Answer

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SA Adler



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