

Write the formulas for each of the following covalent compounds.

1. sulfur tetroxide _____
2. phosphorus trifluoride _____
3. iodine monobromide _____
4. silicon tetrahydride _____
5. selenium dioxide _____
6. oxygen dichloride _____
7. bromine pentachloride _____
8. oxygen difluoride _____
9. phosphorus pentachloride _____
10. nitrogen monoxide _____
11. dinitrogen tetroxide _____
12. diphosphorus pentoxide _____
13. disulfur difluoride _____
14. tetraphosphorus trisulfide _____
15. dinitrogen tetrachloride _____

Name each of the following covalent compounds.

1. CO_2 _____
2. SO_3 _____
3. N_2O_3 _____
4. N_2O_5 _____
5. PCl_3 _____
6. NH_3 _____

Worksheet Molecular Compounds Chemistry A Study Of Matter 6 Key

R Bogdan



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