

## **Building Models of Molecules**

- Different atoms have a different number of holes to represent the different number of connections that they must make with other atoms.
  - o ALL HOLES MUST BE FILLED!!
- Atoms are connected by bonds represented by the connectors
  - o Sometimes, 2 or 3 connectors must be used to fill all the holes

Hydrogen ( ) has \_\_\_\_\_ holes, and must make \_\_\_\_\_ bonds.  
 Oxygen ( ) has \_\_\_\_\_ holes, and must make \_\_\_\_\_ bonds.  
 Nitrogen ( ) has \_\_\_\_\_ holes, and must make \_\_\_\_\_ bonds.  
 Carbon ( ) has \_\_\_\_\_ holes, and must make \_\_\_\_\_ bonds.  
 Chlorine ( ) has \_\_\_\_\_ holes, and must make \_\_\_\_\_ bonds.

Your task: Build the following molecules and complete the chart by drawing their structural diagram

| Molecule/<br>Compound Name | Chemical<br>FORMULA           | STRUCTURAL DIAGRAM |
|----------------------------|-------------------------------|--------------------|
| Hydrogen Gas               | H <sub>2</sub>                |                    |
| Oxygen Gas                 | O <sub>2</sub>                |                    |
| Nitrogen Gas               | N <sub>2</sub>                |                    |
| Methane                    | CH <sub>4</sub>               |                    |
| Water                      | H <sub>2</sub> O              |                    |
| Carbon Dioxide             | CO <sub>2</sub>               |                    |
| Ethane                     | C <sub>2</sub> H <sub>6</sub> |                    |
| Ethylene                   | C <sub>2</sub> H <sub>4</sub> |                    |
| Acetylene                  | C <sub>2</sub> H <sub>2</sub> |                    |

# Worksheet Molecular Compounds Chemistry A Study Of Matter Key

**Christian G. Meyer**



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