

Unit 3 Cellular Energetics AP Biology Exam Review

Enzyme Structure

- 1) To the right is an enzyme-substrate complex. Label what A, B, and C represent.
- 2) Describe how enzymes work. Make sure to include the terms products, reactants, and activation energy.

The activation energy is the energy required to start a reaction. Enzymes are proteins that bind to a molecule, or substrate, to modify it and lower the energy required to make it react. Reactants (or 'substrates') are the starting materials for a reaction. ... Because they speed up reactions, they are called catalysts.



Enzyme Catalysis and Environmental Effects

- 3) How can a scientist make an enzyme reaction increase in production rates?

Enzyme concentration: Increasing enzyme concentration will speed up the reaction, as long as there is substrate available to bind to. ... Substrate concentration: Increasing substrate concentration also increases the rate of reaction to a certain point.

- 4) What could slow down the creation of products from an enzymatic reaction? Give at least three items.

Temperature: Raising temperature generally speeds up a reaction, and lowering temperature slows down a reaction.

- 5) What is denaturation? What causes this process to occur?

If a protein loses its shape, it ceases to perform that function. The process that causes a protein to lose its shape is known as denaturation. Denaturation is usually caused by external stress on the protein, such as solvents, inorganic salts, exposure to acids or bases, and by heat.

pH Math

- 4) What's the concentration of a HCl solution that has a pH of 3.5?

Cell Energy: First and Second Law of Thermodynamics

- 5) Compare the first law to the second law of thermodynamics. Give examples of each.

The first law, also known as Law of Conservation of Energy, states that energy cannot be created or destroyed in an isolated system. The second law of thermodynamics states that the entropy of any isolated system always increases.

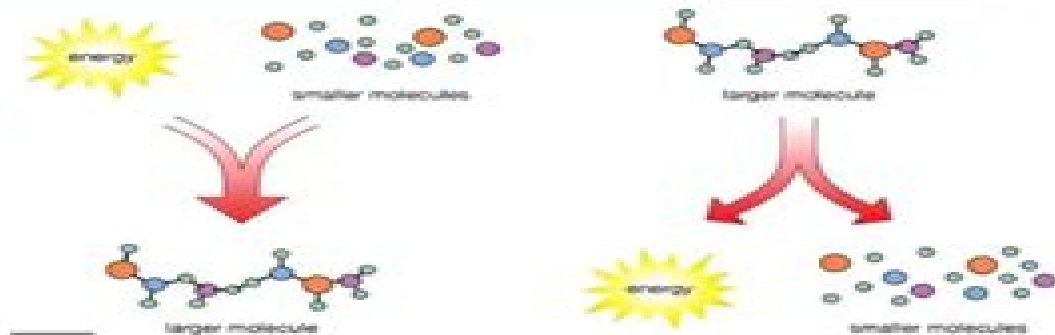
$$\text{pH} = -\log[\text{H}^+]$$
$$[\text{H}^+] = 10^{-\text{pH}}$$

Cell Energy: Catabolic and Anabolic Reactions

- 6) Compare catabolic to anabolic reactions.

Anabolism refers to the process which builds molecules the body needs; it usually requires energy for completion. Catabolism refers to the process that breaks down complex molecules into smaller molecules; it usually releases energy for the organism to use.

- 7) Look at the picture to the right. Which diagram (A or B) is an anabolic reaction and which one is a catabolic reaction. How can you tell?



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Robinson



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