

Chapter 7 Chemical Reactions

6. What is a double-replacement reaction? A double-replacement reaction is a reaction in which two different compounds exchange positive ions and form two new compounds.
7. Complete the chart by filling in the general forms of the reactions shown.

General Forms	
Single-Replacement Reaction	Double-Replacement Reaction
$A + BC \longrightarrow B + AC$	$AB + CD \longrightarrow AD + CB$

8. Lead(II) nitrate reacts with potassium iodide to form lead(II) iodide and potassium nitrate. Write the balanced equation for this double-replacement reaction. $Pb(NO_3)_2 + 2KI \longrightarrow PbI_2 + 2KNO_3$
9. Circle the letter of the correct answer. Calcium carbonate, $CaCO_3$, reacts with hydrochloric acid, HCl , in a double-replacement reaction. What are the products of this reaction?
- ☒ a. calcium chloride, $CaCl_2$, and carbonic acid, H_2CO_3
 b. calcium hydride, CaH_2 , chlorine, Cl_2 , and carbon dioxide, CO_2
 c. calcium hydrogen carbonate, $Ca(HCO_3)_2$, and chlorine, Cl_2
 d. calcium perchlorate, $Ca(ClO_4)_2$, and methane, CH_4
10. Is the following sentence true or false? A combustion reaction is a reaction in which a substance reacts with carbon dioxide, often producing heat and light. false
11. Methane, CH_4 , burns in oxygen to form carbon dioxide and water. Write the balanced equation for this reaction. $CH_4 + 2O_2 \longrightarrow CO_2 + 2H_2O$
12. Is the following sentence true or false? The reaction that forms water can be classified as either a synthesis reaction or a combustion reaction. true

Reactions as Electron Transfers (pages 204–205)

13. What is an oxidation-reduction reaction? An oxidation-reduction reaction is a reaction in which electrons are transferred from one reactant to another.
14. Calcium reacts with oxygen to form calcium oxide. Which reactant is oxidized in this reaction? calcium
15. Is the following sentence true or false? When calcium reacts with oxygen, each calcium atom gains two electrons and becomes a calcium ion with a charge of 2–. false
16. Is the following sentence true or false? Oxygen must be present in order for an oxidation-reduction reaction to take place. false
17. The process in which an element gains electrons during a chemical reaction is called reduction.

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Jake Mariska



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Balancing Chemical Equations Worksheets (Over 200 Reactions to Balance) Chris McMullen, 2016-01-12 Master the art of balancing chemical reactions through examples and practice 10 examples are fully solved step by step with explanations to serve as a guide Over 200 chemical equations provide ample practice Exercises start out easy and grow progressively more challenging and involved Answers to every problem are tabulated at the back of the book A chapter of pre balancing exercises helps develop essential counting skills Opening chapter reviews pertinent concepts and ideas Not just for students Anyone who enjoys math and science puzzles can enjoy the challenge of balancing these chemical reactions

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numerical problems in book An updated solutions manual and PowerPoint slides are also provided in addition to new video guides and UniSim program files **Publications** United States. National Bureau of Standards,1982 Publications of the National Institute of Standards and Technology ... Catalog National Institute of Standards and Technology (U.S.),1982

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