## WORD EQUATIONS

Name <u>Key</u>

write the world equations below as chemical equations and balance.

1. zinc + lead (ii) nitrate yield zinc nitrate + lead

 $Z_n + P6(NO_3)_2 \rightarrow Z_n(NO_3)_2 + P6$ 

2. aluminum bromide + chlorine yield aluminum chloride + bromine

2A1Br, +3C/2 →2A1C/3+3Br2

 sodium phosphate + coloium chloride yield calcium phosphate + sodium chloride

 $ZNL_{1}, PO_{V(ng)} + 3CaCl_{2(ng)} \longrightarrow Ca_{1}(Ry)_{2(n)} + 6NhCl_{ng}$ 

4. potassium chiarate when heated yields potassium chiaride + axygen gas

2 KC103 - →2KC1 +302

5. aluminum + hydrochloric acid yield aluminum chloride + hydrogen gas

2A1 +6HC1 -> 2A1C1, +3H2

6. calcium hydroxide + phasphoric acid yield calcium phasphate + water

3 Ca (OH)<sub>2 (mg)</sub> +2H, PO<sub>4 (g)</sub> → Ca, (PO<sub>4</sub>)<sub>2g</sub> +6H<sub>2</sub>O<sub>g</sub> 5/44+4 506666

7. oppper + sulfuric acid yield capper (ii) sulfate + water + sulfur dioxide

Cu + 2H2504 -> Cu504+2H20 + 502

hydrogen + nitrogen monoxide yield water + nitrogen

 $2H_2 + 2NO \rightarrow 2H_2O + N_3$ 

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**WJ Hussar** 

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