

Chapter 6 Chemical Bonds

Word Wise

Use the terms from the following list to fit each of the clues given below.

molecular	covalent	polyatomic
crystals	anion	polar
ionic	alloy	metallic
formula	bond	cation

Clues

Vocabulary Terms

A type of bond that holds cations and anions together

A type of bond in which two atoms share a pair of valence electrons

A neutral group of atoms that are joined together by one or more covalent bonds

A term describing a covalent bond in which electrons are not shared equally

An ion that contains a covalently bonded group of atoms

An ion with a negative charge

An ion with a positive charge

A notation that shows what elements a compound contains and the ratio of the atoms or ions of these elements in the compound

Solids whose particles are arranged in a lattice structure

A mixture of two or more elements, at least one of which is a metal

A type of bond that exists between a metal cation and the shared electrons that surround it

The force that holds atoms or ions together

If you finish, try solving a few extra at the end.

Wordwise Chpater 6 Chemical Bonds

Robert Thomas Sanderson



Wordwise Chapter 6 Chemical Bonds:

Prentice Hall Physical Science Concepts in Action Program Planner National Chemistry Physics Earth Science, 2003-11 Prentice Hall Physical Science Concepts in Action helps students make the important connection between the science they read and what they experience every day Relevant content lively explorations and a wealth of hands on activities take students understanding of science beyond the page and into the world around them Now includes even more technology tools and activities to support differentiated instruction *Chemical Bonds* Phillip Manning, 2009 Modern life is made up of a mind boggling array of materials A simple drinking cup for example might be made of Styrofoam paper or glass depending on the drinkers needs at the moment Home storage cabinets can be made of metal wood or plastic Space shuttles are assembled from silicon steel and hundreds of other materials All of these items owe their properties to the chemical bonds between the atoms that make up the substance *Chemical Bonds* examines the nature of the chemical bonds answering fundamental questions about how they form how they are broken and how they help define life as we know it *Chemical Bonds* Jeremy K. Burdett, 1997-05-21 Taking a question and answer format this tutorial explores the fundamental concepts of chemical bonding and is based on the lectures of Professor Jeremy Burdett **Chemical Bonding Clarified Through Quantum Mechanics** George C. Pimentel, Richard D. Spratley, 1969 **Chemical Bonds and Bond Energy** Robert Thomas Sanderson, 1971

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