

Word equations

Write the **skeleton** equation for each of the following reactions.

1. hydrogen + oxygen → water



2. iron(III) oxide + hydrogen → water + iron



3. sodium + water → sodium hydroxide + hydrogen



4. calcium carbide + oxygen → calcium + carbon dioxide



5. potassium iodide + chlorine → potassium chloride + iodine



6. chromium + tin(IV) chloride → chromium(III) chloride + tin



7. magnesium + copper(II) sulphate → magnesium sulphate + copper



8. zinc sulphate + strontium chloride → zinc chloride + strontium sulphate



9. ammonium chloride + lead(II) nitrate → ammonium nitrate + lead(II) chloride



10. iron(III) nitrate + magnesium sulphide → iron(III) sulphide + magnesium nitrate



11. aluminum chloride + sodium carbonate → aluminum carbonate + sodium chloride

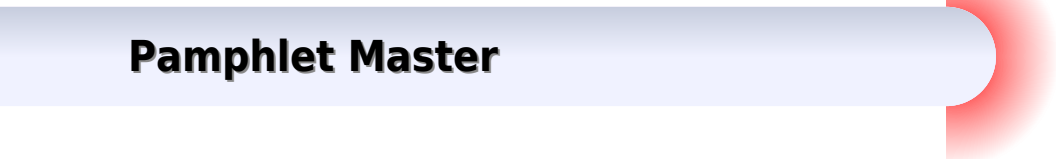


12. sodium phosphate + calcium hydroxide → sodium hydroxide + calcium phosphate



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Chemistry Homework Frank Schaffer Publications, Joan DiStasio, 1996-03 Includes the periodic table writing formulas balancing equations stoichiometry problems and more **Coordinating Procedural and Conceptual Knowledge to Make Sense of Word Equations: Understanding the Complexity of a 'simple' Chemical Task at the Learner's Resolution** Keith Stephen Taber, Pat Bricheno, 2009 Abstract This paper discusses the conceptual demands of an apparently straightforward task set to secondary level students completing chemical word equations with a single omitted term Chemical equations are of considerable importance in chemistry and school students are expected to learn to be able to write and interpret them However it is recognized that many students find them challenging The present paper explores students accounts of their attempts to identify the missing terms to illuminate why working with chemical word equations is so challenging from the learner's perspective 300 secondary age students responded to a 5 item exercise based on chemicals and types of reactions commonly met at school level For each item they were asked to identify the missing term in a word equation and explain their answers This provided a database containing more than a thousand student accounts of their rationales Analysis of the data led to the identification of seven main classes of Word Equations and Related Topics INTERNATIONAL WORKSHOP ON WORD EQUATIONS AND RELATED TOPICS, 1993 Chemical Equations Lifeliqe, 2019 This lesson plan covers describing chemical reactions using word equations knowing the correct symbols to use to write skeleton equations for chemical reactions and balancing chemical equations *Balancing Chemical Equations* Learning Hub Publishing, 2020-06-22 To improve their chemistry one must perfect the skill of balancing numerous chemical equations and doing them properly This is something which you will have to practice again and again to perfect as its a key skill when it comes to high school and even in college entrance exam and some college courses In this book you will find multiple practice equations with answers in 3 terms 4 terms 5 terms and many word problems that you can work on and improve your chemistry How to Solve Word Problems in Chemistry David E. Goldberg, 2001-07-17 In addition to having to master a vast number of difficult concepts and lab procedures high school chemistry students must also learn with little or no coaching from their teachers how to solve tough word problems Picking up where standard chemistry texts leave off How to Solve Word Problems in Chemistry takes the fear and frustration out of chemistry word problems by providing students with easy to follow procedures for solving problems in everything from radioactive half life to oxidation reduction reactions *Word Equations and Related Topics* Klaus U. Schulz, 2014-01-15 *Chemical Reactions and Their Equations* Ingo Waldemar Dagobert Hackh, 1921 **Word Equations with One and Two Unknowns** Robert Dąbrowski, 2005 *Balancing Chemical Equations Workbook* Kraaya Publishing, 2021-02-21 Chemical Reactions to Balance Workbook This chemistry balancing equations practice workbook contains 250 non balanced chemical equations Begin with 2 terms problems Work your way up to 6 terms problems This is the perfect workbook to increase chemistry balancing skills for beginners Table of contents How

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