

Name : _____

Writing Balanced Chemical Equations

Write balanced chemical equations for the following reactions:

- (a) aluminum bromide + chlorine $\rightarrow$ aluminum chloride + bromine
- (b) calcium sulfite $\rightarrow$ calcium oxide + sulfur dioxide
- (c) phosphoric acid + calcium hydroxide $\rightarrow$ calcium phosphate + water
- (d) silver nitrate + ammonium carbonate $\rightarrow$ silver carbonate + ammonium nitrate
- (e) lithium hydrogen carbonate $\rightarrow$ lithium carbonate + water + carbon dioxide
- (f) magnesium + silver acetate $\rightarrow$ magnesium acetate + silver
- (g) aluminum bromide + potassium sulfate $\rightarrow$ potassium bromide + aluminum sulfate
- (h) sodium chloride + fluorine $\rightarrow$ sodium fluoride + chlorine
- (i) copper + sulfuric acid $\rightarrow$ copper sulfate + water + oxygen
- (j) zinc + iron (III) chloride $\rightarrow$ zinc chloride + iron

Word Equations To Balance Chemical Equations

Graeme Lofts, Merrin J. Evergreen



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Chemistry Equations & Answers Speedy Publishing, 2014-08-01 One of the most important parts of learning chemistry is simply knowing all of the equations and formulas that are used in it It is important to know the exact formatting of these equations and most classes will require a student to know them for exams It is a good idea to learn these equations and formulas with the use of a study pamphlet The pamphlet can condense all of the information so a student can memorize the equations and formulas while studying

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Objective Workbook for Simplified Middle School Chemistry ,

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Foundations of College Chemistry Morris Hein, Susan Arena, Cary Willard, 2023 Foundations of College Chemistry 16th edition presents chemistry as a modern vital subject and is designed to make introductory chemistry accessible to all beginning students It is intended for students who have never taken a chemistry course or those who had a significant interruption in their studies but plan to continue with the general chemistry sequence The central focus is to make chemistry interesting and understandable and teach students the problem solving skills they will need This International Adaptation offers new and updated content with improved presentation of all course material It builds on the strengths of previous editions including clear explanations and step by

step problem solving The material emphasizes real world applications of chemistry as the authors develop the principles that form the foundation for the further study of chemistry There is new and expanded coverage of polarizing power and polarizability Fajans rules collision number and mean free path abnormal molecular masses and van t Hoff factor and applications of radioactivity Learning Chemistry 7 Solution Book (Year 2023-24) ,2024-01-02 **Introduction to**

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Lofts,2025-12-03 **Jacaranda Science 10 for Western Australia, 5e LearnON and Print** Jacaranda,2025-10-10

Arun Deep's ICSE Formula Handbook Physics, Chemistry, Mathematics For Classes 9 and 10 Vijay Garg,2025-04-01 This comprehensive book incorporates chapterwise formulas for ICSE Physics Chemistry and Mathematics It encompasses all the essential formulas to facilitate quick revision for students Representing a detailed and valuable resource this book stands as an elaborate and highly useful guide for students **Oswaal NCERT One for All Book For UPSC and State PSC's**

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