

Chapter 10 Chemical Quantities Review

Part 1: Vocabulary

Label the following descriptions with the correct vocabulary word.

- | | |
|-------------------------|---|
| _____ mass | _____ percent composition. |
| _____ empirical formula | _____ empirical formula |
| _____ molar mass | _____ standard temperature and pressure (STP, 1 atm, 0°C) |

1. _____ is a description of the relative amounts of each element in a compound.
2. _____ is the lowest whole-number ratio of the atoms of the elements in a compound.
3. _____ is the lowest structural formula and mass of any gas at STP.
4. _____ is the lowest structural formula of a compound.
5. _____ is the lowest structural formula of a compound.
6. _____ is the lowest structural formula of a compound.
7. _____ is the lowest structural formula of a compound.
8. _____ is the lowest structural formula of a compound.
9. _____ is the lowest structural formula of a compound.
10. _____ is the lowest structural formula of a compound.

Part 2: Section Concept Review

10.1 The Mole: A Measurement

Complete the following statement by choosing the correct word or phrase. Consider the following concepts: mass, volume, number of particles, and amount of substance.

1. _____ is the mass of a substance, measured in grams, and is related to a specific quantity called the _____.
2. _____ is the volume of a substance, measured in liters, and is related to a specific quantity called the _____.
3. _____ is the number of particles of a substance, measured in moles, and is related to a specific quantity called the _____.
4. _____ is the amount of substance, measured in moles, and is related to a specific quantity called the _____.
5. _____ is the number of particles of a substance, measured in moles, and is related to a specific quantity called the _____.

Practice Problems

1. _____ is the mass of a substance, measured in grams, and is related to a specific quantity called the _____.
2. _____ is the volume of a substance, measured in liters, and is related to a specific quantity called the _____.
3. _____ is the number of particles of a substance, measured in moles, and is related to a specific quantity called the _____.
4. _____ is the amount of substance, measured in moles, and is related to a specific quantity called the _____.
5. _____ is the number of particles of a substance, measured in moles, and is related to a specific quantity called the _____.

Problems

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Ying Liu



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