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## Unit 6 – Worksheet 4 Molecular Compounds

1

Properties.

Basic structural unit

- Name each of the following binary compounds of non-metallic elements
- a CBr4
- b. N<sub>2</sub>P<sub>3</sub>
- c. PCl<sub>3</sub>
- d: ICI
- e. N2<sup>O</sup>
- f SiF4
- g GeH<sub>4</sub>
- h N<sub>2</sub>Br<sub>4</sub>
- i. P<sub>2</sub>S<sub>5</sub>
- j SeO2
- k NH3
- 1.802

- Write the formula for the following binary compounds of nonmetallic elements
- a. phosphorus triiodide
- b. silicon tetrachloride
- c. dinitrogen pentoxide
- e dinitrogen tetroxide
- f. carbon monoxide
- g carbon dioxide
- h sulfur hexafluoride
- i dinitrogen tetrachloride
- j. carbon tetrasodide
- k phosphorus pentafluoride
- 1 diphosphorus pentoxide

## **Unit 6 4 Molecular Compounds Answers**

**PT Brinkman** 

Unit	6	4	Molecular	Compounds	<b>Answers:</b>
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