

Chemistry Unit 8

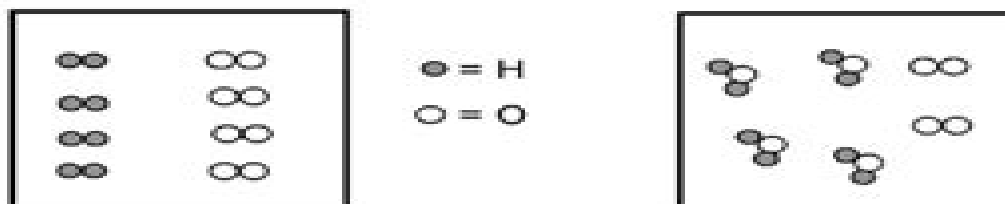
Worksheet 3: Adjusting to Reality - Limiting Reactant

1. Write the balanced equation for the reaction between hydrogen and oxygen.

Balanced Equation: 2 $\text{H}_{2(g)}$ + 1 $\text{O}_{2(g)}$ $\rightarrow$ 2 $\text{H}_2\text{O}_{(g)}$

Suppose that 4 molecules of hydrogen gas and 4 molecules of oxygen gas react to form water.

Make a drawing that represents the reaction container before and after the reaction.



Before

After

4 molecules How many molecules of water can be produced?

oxygen Which reactant is in excess? Why? There are leftover molecules

2 molecules How many molecules of excess reactant are there?

Construct a Before-Change-After Table for this reactant mixture:

In this table the numbers refer to molecules rather than moles.

Equation: 2 $\text{H}_{2(g)}$ + 1 $\text{O}_{2(g)}$ $\rightarrow$ 2 $\text{H}_2\text{O}_{(g)}$

Before 4 4 0

Change -4 -2 +4

After 0 2 4

According to the table you just made,

4 molecules How many molecules of water can be produced?

oxygen Which reactant is in excess? Why? 2 molecules are left-over "after" the reaction

2 molecules How many molecules of excess reactant are there?

Based on your two methods of analysis above, what determines how much product can be made from a particular reactant mix?

The determining factor is which of the reactants runs out first. This determines how much product can be made, even if another reactant is in excess.

Unit 7 3 Limiting Reactant Answers

Arturo Cuomo



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Basic Concepts of Chemistry Leo J. Malone, Theodore Dolter, 2008-12-03 Engineers who need to have a better understanding of chemistry will benefit from this accessible book It places a stronger emphasis on outcomes assessment which is the driving force for many of the new features Each section focuses on the development and assessment of one or two specific objectives Within each section a specific objective is included an anticipatory set to orient the reader content discussion from established authors and guided practice problems for relevant objectives These features are followed by a set of independent practice problems The expanded Making it Real feature showcases topics of current interest relating to the subject at hand such as chemical forensics and more medical related topics Numerous worked examples in the text now include Analysis and Synthesis sections which allow engineers to explore concepts in greater depth and discuss outside relevance

Chemistry Questions - Answers for Intermediate Mr. Ajeet Singh, 2018-05-13 A question answer formatted book designed for intermediate level chemistry students covering theory based and numerical questions to aid learning and exam preparation

Conceptual Chemistry Volume I For Class XI S.K. Jain & Shailesh K. Jain, 1998 Conceptual Chemistry Volume I For Class XI

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