# Stoichiometry "End of Chapter" IN CLASS Practice Assessment

thackground into -

Why are the coefficients in a balanced equation necessary for dischiometry?
 Coefficients are necessary for establishing a male male ratio to convert between modes of different substances.

Skills - Using a balanced equation, calculate using mole to mole ratics & Using a balanced equation, given the grams of reactant, calc. the grams of product

2. Consider the following unbalanced equation:

4 HCl + MnO, -> 2 H<sub>2</sub>O + MnCl<sub>2</sub> + Cl<sub>2</sub>

There are 5.00g of manganese(N) oxide present, calculate the grame of chiorine gas. 5.00g MnO<sub>2</sub> × 1mol MnO<sub>2</sub> × 1mol Cl<sub>2</sub> × 70.9gCl<sub>3</sub> = 4.00g Cl<sub>3</sub> = 600 g Cl<sub>3</sub> = 600

Skill - After calculating theoretical yield in grams, calculate the percent yield.

You are assigned to complete a lab. Before the reaction is complete, you calculate that a
reaction should produce 4 0g Culpi. After the lab is over you find that 2.5g Culpi was
collected. What would the percent yield be for that reaction completed within the lab?
No yield — Lactual/Theoretical() x 2.00
 — (2.0ex), 0ex) x 1.00 — 200s.

Skill - Given experimental data, determine the limiting reactant and the excess reactant

4. Calcium cartionate continues with gaseous hydrogen chloride to produce solid calcium chloride, gaseous carbon disolde, and water vapor. What is the percent yield if 155g of calcium carbonate combines with 250 0g of hydrogen chloride and only 342 0g of calcium chloride in actually obtained within the list setting? Complete and show work for all 5 steps of a stotchomotry problem.

CaCO<sub>3</sub> + 2 HCl → CaCl<sub>3</sub> + CO<sub>3</sub> + H<sub>2</sub>O

130<sub>9</sub> 142,9 actual

155g CaCO<sub>2</sub> x 1mol CaCO<sub>2</sub>/100.09g CaCO<sub>2</sub> ~ 1.55mol CaCO<sub>2</sub> /1mol in non = 1.55 nons 250a HCL x 1mol HCl/36.458g HCl = 6.057 mol HCl /2mol in non = 3.43 nons

1.55mol CaCO,  $\times$  (1mol CaCly1mol CaCO)  $\times$  (110.90gCaCly1mol CaCly) = 172.02n

%yield = [actual/theoretical] x 100 = [142.0g/172.02g] x 100 = 82.55%

# **Unit 7 Stoichiometry Review**

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#### **Unit 7 Stoichiometry Review:**

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further as the years go by beyond the reach of our present imagination and new research applications will continue to emerge This book therefore is not intended to be a comprehensive review or textbook on SPM Its aim is to cover only a selected part of the active re search fields of SPM and related topics in which I have been directly involved over the years These include the basic principles of STM and AFM and their applications to fullerene film growth SiC surface reconstructions MBE molecular beam epitaxy growth of CaAs atomic scale manipulation of Si surfaces and meso scopic work Comprehensive Energy Systems Ibrahim Dincer, 2018-02-07 Comprehensive Energy Systems Seven Volume Set provides a unified source of information covering the entire spectrum of energy one of the most significant issues humanity has to face This comprehensive book describes traditional and novel energy systems from single generation to multi generation also covering theory and applications In addition it also presents high level coverage on energy policies strategies environmental impacts and sustainable development No other published work covers such breadth of topics in similar depth High level sections include Energy Fundamentals Energy Materials Energy Production Energy Conversion and Energy Management Offers the most comprehensive resource available on the topic of energy systems Presents an authoritative resource authored and edited by leading experts in the field Consolidates information currently scattered in publications from different research fields engineering as well as physics chemistry environmental sciences and economics thus ensuring a common standard and language Cehmistry Textbook for College and University USA Ibrahim Sikder, 2023-06-04 Cehmistry Textbook USA Chemistry II For Dummies John T. Moore, 2012-06-08 The tools you need to ace your Chemisty II course College success for virtually all science computing engineering and premedical majors depends in part on passing chemistry. The skills learned in chemistry courses are applicable to a number of fields and chemistry courses are essential to students who are studying to become nurses doctors pharmacists clinical technicians engineers and many more among the fastest growing professions But if you re like a lot of students who are confused by chemistry it can seem like a daunting task to tackle the subject That's where Chemistry II For Dummies can help Here you'll get plain English easy to understand explanations of everything you ll encounter in your Chemistry II class Whether chemistry is your chosen area of study a degree requirement or an elective you ll get the skills and confidence to score high and enhance your understanding of this often intimidating subject So what are you waiting for Presents straightforward information on complex concepts Tracks to a typical Chemistry II course Serves as an excellent supplement to classroom learning Helps you understand difficult subject matter with confidence and ease Packed with approachable information and plenty of practice opportunities Chemistry II For Dummies is just what you need to make the grade Cumulated Index Medicus ,1987 5 Steps to a 5: AP Chemistry 2021 Mary Millhollon, Richard H. Langley, 2020-10-02 MATCHES THE LATEST EXAM In this hybrid year let us supplement your AP classroom experience with this multi platform study guide The immensely popular 5 Steps to a 5 AP Chemistry guide has been updated for the 2020 21 school year and now contains 3 full length practice exams available both in the book and online that reflect the latest exam Up to Date Resources for COVID 19 Exam Disruption Access to a robust online platform Comprehensive overview of the AP Chemistry exam format Hundreds of practice exercises with thorough answer explanations Proven strategies specific to each section of the test A self guided study plan including flashcards games and more online **Pp/Chemistry** BarCharts, Inc.,2008-06-18 <u>CHEMISTRY</u> SILBERBERG,2003

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