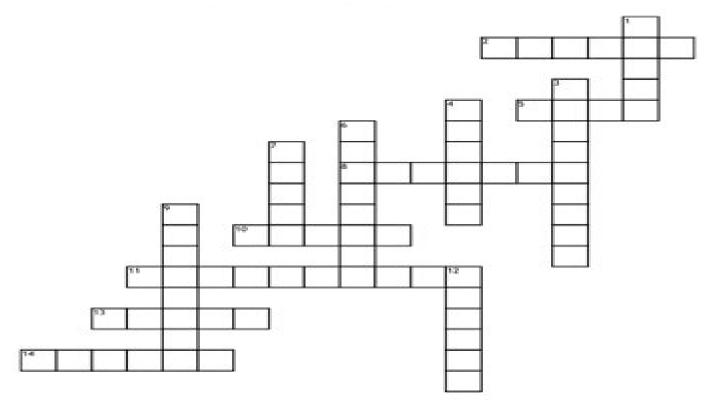
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Across

- 2. atom with positive charge
- 5. number of electrons shared in a double bond
- 8. electrons in highest energy level
- 10. rule that atoms want 8 electrons in outer shell

- free ions dissociate and can carry electricity
- 13. atom with negative charge
- 14. melting point of ionic compounds

Down

- melting point of covalent compounds
- 3. combine to make covalent bonds

- 4. oxidation number of Phosphorous
- bond where charges are shared
- bond where electrons are transferred
- charges that attract
- 12. sodium's atomic
- number

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