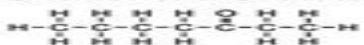


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1. Which of the following shows the correct condensed structural formula of a compound that has five carbon (C) atoms, 12 hydrogen (H) atoms and one oxygen (O) atom?

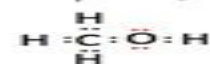
- ☐ $\text{CH}_3-\text{C}(\text{CH}_2)_2-\text{CH}_2-\text{C}(=\text{O})-\text{H}$
- ☐ $\text{CH}_3-\text{C}(\text{CH}_3)(\text{CH}_2)-\text{CH}_2-\text{C}(=\text{O})-\text{H}$
- ☐ $\text{CH}_3-\text{CH}(\text{CH}_3)-\text{CH}(\text{CH}_3)-\text{CH}_2-\text{O}$
- ☐ $\text{CH}_3-\text{C}(\text{CH}_3)_2-\text{CH}_2-\text{C}(=\text{O})-\text{H}$

2. Which of the following is the correct skeletal structural formula for the following compound?



- ☐
- ☐
- ☐
- ☐

3. What is the correct line-bond structural formula for the compound shown below? (The black dots are bonding electrons between the atoms and the red dots are nonbonding electrons.) Hint: Since this is an organic compound, nonbonding electrons are, in general, not shown in the line-bond structural formula



- ☐ $\text{H}-\text{C}(\text{H})_2-\text{C}(=\text{O})-\text{H}$
- ☐ $\text{H}-\text{C}(\text{H})_2-\text{O}-\text{H}$
- ☐
- ☐ $\text{CH}_3-\text{O}-\text{H}$

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