

## DEPARTMENT OF CHEMISTRY

College of Engineering and Technology SRM Institute of Science and Technology 16 petition is admirissing -- 600,32003. Internal Assessment - II

Program: B.Tech Course Code & Title: 21CVB101J/Chemistry Year & Sem: I/II

1 hater 29-03-2023 Times 12.30-1.30 pm Minu. Minutes: 30 Minutes.

## Answer ALL, the MCOs Part-A (10\*1-10 Marks)

- 1. Which of the following is not a state function
  - a) AG b) AE c) W d) H
- 2. Internal energy does not include
  - a) Nuclear energy b) Vibrational energy c) Rotational energy d) Energy of gravitational pull
- 3. Gibbs energy change AG is related to equilibrium constant K as:
  - a)  $\Delta G^{\circ} = -RT \ln k$  b)  $\Delta G^{\circ} = RT \ln k$  c)  $\ln k = -RT/\Delta G^{\circ}$  d)  $\ln k = \Delta G^{\circ}/RT$
- The conductivity of electrolytic conductors is due to \_\_\_\_
  - a) Plow of free mobile electrons b) Movement of ions c) Either movement of electrons or ions d) Cannot be said
- 5. The process of deterioration of a metal due to unwanted chemical or electrochemical
- interaction of the metal with its environment is called \_ a) Electrolysis b) Electrodialysis c) Corrosion d) Deposition
- 6. Chemical used in salt bridge is
  - a) KOH b) KCI c) KNO2 d) KBr
- 7. If a compound has 3 chiral carbons What is the number of optically active isomers?
  - a) 9 b) 3 c) 4 d) 8
- The potential energy of n-butane is minimum for
  - a) Skew conformations b) Staggered conformations c) Eclipsed conformations
  - d) Gauche
- Optical isomerism is a type of \_
  - a) metamerism b) stereoisomerism c) geometrical isomerism d) tautomerism

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