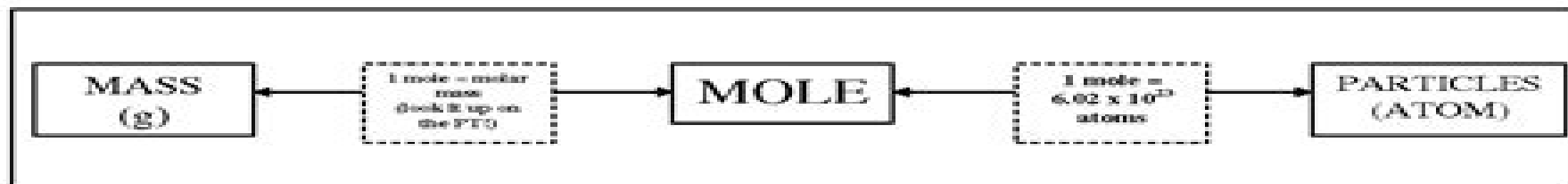


1 mole = 6.02×10^{23} atoms 1 mole = atomic mass (g)



Two Step Problems:

1. What is the mass of 2.25×10^{25} atoms of lead?

$$\frac{2.25 \times 10^{25} \text{ atoms of lead}}{6.02 \times 10^{23} \text{ atoms}} \times \frac{1 \text{ mole}}{1 \text{ mole}} \times \frac{207.2 \text{ g}}{1 \text{ mole}} = 7744.19 \text{ g} = 7740 \text{ g}$$

2. How many atoms are in 10.0g of gold?

$$\frac{10 \text{ g gold}}{197.0 \text{ g}} \times \frac{1 \text{ mole}}{1 \text{ mole}} \times \frac{6.02 \times 10^{23} \text{ atoms}}{1 \text{ mole}} = 3.06 \times 10^{22} \text{ atoms}$$

PRACTICE PROBLEMS:

1. How many moles are equal to 625g of copper?

$$\frac{625 \text{ g of copper}}{63.5 \text{ g Cu}} \times \frac{1 \text{ mole}}{1 \text{ mole}} = 9.77 \text{ mol Cu}$$

2. How many moles of barium are in a sample containing 4.25×10^{23} atoms of barium?

$$\frac{4.25 \times 10^{23} \text{ atoms of barium}}{6.02 \times 10^{23} \text{ atoms}} \times \frac{1 \text{ mole}}{1 \text{ mole}} = 0.706 \text{ mol}$$

3. Convert 2.35 moles of carbon to atoms.

$$\frac{2.35 \text{ moles}}{1 \text{ mole}} \times \frac{6.02 \times 10^{23} \text{ atoms}}{1 \text{ mole}} = 1.42 \times 10^{24} \text{ atoms}$$

4. How many atoms are in 4.0g of potassium?

$$\frac{4.0 \text{ g}}{39.1 \text{ g}} \times \frac{1 \text{ mole}}{1 \text{ mole}} \times \frac{6.02 \times 10^{23} \text{ atoms}}{1 \text{ mole}} = 6.2 \times 10^{22} \text{ atoms}$$

Word Mix Mole Equation And Key

SB Merriam



Word Mix Mole Equation And Key:

Word Mix Mole Equation And Key Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become much more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Word Mix Mole Equation And Key**," published by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve to the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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