

# Workbook Answer Key

## UNIT 1

- 1 A. Hello. What's your name?  
B. Suzanne. What's your name?  
A. My name is John. Where are you from, Suzanne?  
B. I'm from London. Where are you from?  
A. I'm from New York.
- 2 3. I'm from Russia. 4. I'm from Japan.  
5. We're from Brazil. 6. I'm from Taiwan. 7. We're from Australia.
- 3 1. She's Tomoko Suzuki. 2. She's from Japan. 3. She's a student. 4. She's 19.  
5. They're Max and Lisa Smith.  
6. They're from the United States.  
7. They are teachers. 8. Lisa is 42.  
9. Max is 45.
- 4 2. What is your name? 3. She is from France. 4. I am ten years old.  
5. They are from England. 6. You are American. 7. We are students.
- 5 2. My name's Maria. 3. He's married.  
4. They're from Brazil. 5. I'm from Mexico. I'm not from Spain. 6. We're from the United States. 7. He's a teacher. He's not a doctor.

- 6 1. My 2. your 3. your 4. my  
7 1. her 2. His 3. His 4. Her

| Country           | Nationality |
|-------------------|-------------|
| Australia         | Australian  |
| Brazil            | Brazilian   |
| Canada            | Canadian    |
| France            | French      |
| Italy             | Italian     |
| Japan             | Japanese    |
| Mexico            | Mexican     |
| the United States | American    |
| Russia            | Russian     |
| Spain             | Spanish     |
| Taiwan            | Taiwanese   |

|   |                                                                                                                       |                                                                                                      |
|---|-----------------------------------------------------------------------------------------------------------------------|------------------------------------------------------------------------------------------------------|
| 9 | <div><br/>Russia<br/>Spanish</div> | <div></div>       |
|   | <div><br/>C</div>                  | <div><br/>D</div> |

- 10 4. a teacher 5. an apple 6. a car  
7. newspapers 8. a house 9. magazines  
10. a camera 11. a telephone 12. stamps

- 11 2. Champagne is a French drink.  
3. Oxford is an English university.  
4. English is an international language.  
5. Sydney is an Australian city.  
6. A Mercedes is a German car.  
7. A Pentax is a Japanese camera.
- 12 3. Where's she from? 4. What's her name? 5. I'm a student. 6. I have two sisters. 7. They're from Japan. 8. It's a Spanish orange. 9. His name's Hiro. 10. Her name is Linda.
- 13 1. one 2. two 3. three 4. four 5. five  
6. six 7. seven 8. eight 9. nine  
10. ten 11. eleven 12. twelve  
13. thirteen 14. fourteen 15. fifteen  
16. sixteen 17. seventeen 18. eighteen  
19. nineteen 20. twenty
- 14 b. nine oranges c. fifteen students  
d. five cars e. six tickets f. three letters  
g. twelve keys h. four postcards  
i. twenty stamps j. two dictionaries

## UNIT 2

- 1 1. d 2. e 3. f 4. a 5. c 6. b  
2 2. How old is he? 3. Where is he now?  
4. What's his address? 5. Where is she now? 6. What's her job? 7. What's her phone number at work? 8. How old are the children? 9. Where are they now?
- 3 Students' own answers
- 4 2. The president of the United States isn't English. He's American.  
3. Snow isn't hot. It's cold.  
4. Five and six isn't twelve. It's eleven.  
5. You aren't English. You're (Italian).  
6. We aren't in a Russian class. We're in an English class.  
7. Rolls-Royce cars aren't cheap. They're expensive.

- 5 3. Yes, he/she is. / No, he/she isn't.  
4. Yes, it is. / No, it isn't.  
5. Yes, it is. / No, it isn't.  
6. Yes, I am. / No, I'm not.  
7. Yes, I am. / No, I'm not.  
8. Yes, they are. / No, they aren't.
- 6 2. It isn't hot today. 3. I'm not married.  
4. Where's the newspaper? 5. They aren't from Washington. 6. We're at work. 7. You're a student.
- 7 2. You are not old. 3. My apartment is small. 4. Her son is not an accountant. 5. His wife is a journalist. 6. They are at work. 7. We are not rich.

- 8 3. is 4. is 5. P 6. is 7. P 8. P 9. is 10. P
- 9 2. Cathy is Louise's sister. 3. Stephen is David's son. 4. Ann is Stephen's aunt. 5. George is Mary's husband. 6. Mary is Cathy's grandmother. 7. David is Cathy's uncle. 8. David is Ann's brother
- 10 Students' own answers
- 11 2. a hot day 3. a young girl 4. an old man 5. an easy exercise 6. a difficult language 7. a fast car 8. an expensive camera
- 12 1. cars 2. classes 3. cities 4. women 5. families 6. addresses 7. people 8. boys 9. sandwiches 10. universities

|      |                                             |                                                   |
|------|---------------------------------------------|---------------------------------------------------|
| 13 2 | one 1<br>ten 10<br>one hundred 100          | two 2<br>twelve 12<br>twenty 20                   |
|      | 3 three<br>13 thirteen<br>30 thirty         | fourteen 14<br>forty-five 45<br>fifty-four 54     |
|      | 5 five<br>15 fifteen<br>50 fifty            | sixteen 16<br>twenty-six 26<br>sixty-one 61       |
|      | 7 seven<br>17 seventeen<br>78 seventy-eight | eighteen 18<br>eighty-seven 87<br>ninety-eight 98 |
|      | 19 nineteen<br>90 ninety<br>99 ninety-nine  | 40 forty<br>65 sixty-five<br>82 eighty-two        |

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5. \$13.75 Thirteen dollars and seventy-five cents.  
6. \$81.00 Eighty-one dollars.  
7. \$125.50 One hundred and twenty-five dollars and fifty cents.

## UNIT 3

She's a flight attendant. She wears a uniform. She travels a lot. She speaks three languages. She serves drinks. She's a doctor. She works in a hospital. She helps sick people. She works with nurses. She wears a white coat. He's a receptionist. He works in a hotel. He answers the phone. He speaks four languages. He works with a computer. He's a mail carrier. He works outside. He walks a lot. He delivers letters. He starts work at 5:00 A.M.

# Workbook Answer Key Nursing Assistant

**Hussin A.Rothana**



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buffers typically consist of an acid base pair with the acid and base differing by the presence or absence of a proton a conjugate acid base pair for instance one of the buffers that maintain the ph of human blood involves carbonic acid  $\text{H}_2\text{CO}_3$  and its conjugate base the bicarbonate ion  $\text{HCO}_3^-$

**acid base reaction wikipedia** - Oct 13 2022

an acid and a base react not to produce a salt and a solvent but to form a new acid and a new base the concept of neutralization is thus absent brønsted lowry acid base behavior is formally independent of any solvent making it more all encompassing than the arrhenius model

14 3 relative strengths of acids and bases chemistry libretexts - Jul 10 2022

oct 4 2023 brønsted lowry acid base chemistry is the transfer of protons thus logic suggests a relation between the relative strengths of conjugate acid base pairs the strength of an acid or base is quantified in its ionization constant  $K_a$  or  $K_b$  which represents the extent of the acid or base ionization reaction

**overview of acids and bases chemistry libretexts** - Jan 16 2023

jan 30 2023 acetic acid is a weak acid  $K_a = 1.8 \times 10^{-5}$  and acetate is a weak base  $K_b = K_w / K_a = 5.6 \times 10^{-10}$  like acids strong and weak bases are classified by the extent of their ionization strong bases disassociate almost or entirely to completion in aqueous solution

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oct 25 2022 table of contents 16 1 brønsted lowry concept of acids and bases 16 2 water and the ph scale 16 3 equilibrium constants for acids and bases 16 4 acid base properties of salts 16 5 acid base salt equilibrium calculations strong acids

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jun 20 2023 the hard soft acid base principle hsb principle stems from the recognition that some lewis acids and bases seem to have a natural affinity for one another consider the following some metals are commonly found in nature as salts of

chloride or as oxide ores while others are found in combination with sulfur

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jul 19 2022 47428 acids and bases are common substances found in many every day items from fruit juices and soft drinks to soap in this chapter we will examine the properties of acids and bases and learn about the chemical nature of these important compounds we will cover ph and how to calculate the ph of a solution

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may 13 2023 to understand the differences between the three definitions of acids and bases identify acids bases and conjugate acid base pairs according to the three definitions of acids and bases to understand the concept of conjugate acid base pairs in acid base reactions

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acids bases and enzymes many acids and bases in living things provide the ph that enzymes need enzymes are biological catalysts that must work effectively for biochemical reactions to occur most enzymes can do their job only at a certain level of acidity cells secrete acids and bases to maintain the proper ph for enzymes to do their work

acids and bases definition examples properties uses with - May 08 2022

acid base titration difference between acid and base acids and bases definition an acid is any hydrogen containing substance that is capable of donating a proton hydrogen ion to another substance a base is a molecule or ion able to accept a hydrogen ion from an acid acidic substances are usually identified by their sour taste

5 1 acids nomenclature chemistry libretexts - Aug 11 2022

a binary acid is an acid that consists of hydrogen and one other element the most common binary acids contain a halogen the acid name begins with the prefix hydro followed by the base name of the anion followed by the suffix ic

**difference between acid and base properties and types** - Mar 06 2022

oct 18 2023 most of the acids are solid it has a sour taste they produce hydrogen after reacting with active metals like zinc magnesium aluminium or iron acids have a low ph value i e below 7 salts and water are produced when an acid reacts with base or alkalis most of the acids are solids

8 acids and bases medicine libretexts - Feb 05 2022

feb 21 2022 both acids and bases are important enough that we devote an entire chapter to them their properties and their reactions 8 1 brønsted lowry definition of acids and bases 8 2 water is both an acid and a base

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jun 12 2019 acids and bases in aqueous solutions will conduct electricity because they contain dissolved ions therefore acids and bases are electrolytes strong acids and bases will be strong electrolytes weak acids and bases will be weak electrolytes

**6 1 what is an acid and a base chemistry libretexts** - Nov 14 2022

arrhenius s definition of acids and bases the earliest definition of acids and bases is arrhenius s definition which states that an acid is a substance that forms hydrogen ions  $H^+$  when dissolved in water and a base is a substance that forms hydroxide ions  $OH^-$  when dissolved in water

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mar 24 2021 acids react with bases to produce a salt compound and water when equal moles of an acid and a base are combined the acid is neutralized by the base the products of this reaction are an ionic compound which is labeled as a salt and water

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the reaction of an acid and a base is called a neutralization reaction although acids and bases have their own unique chemistries the acid and base cancel each other s chemistry to produce a rather innocuous substance water in fact the general reaction between an acid and a base is  $acid + base \rightarrow water + salt$

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may 20 2018 acids react with bases to produce a salt compound and water when equal moles of an acid and a base are combined the acid is neutralized by the base the products of this reaction are an ionic compound which is labeled as a salt and water

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nov 13 2022 suggest simple tests you could carry out to determine if an unknown substance is an acid or a base state the chemical definitions of an acid and a base in terms of their behavior in water write the formula of the salt formed when a given acid and base are combined