

Name: _____ Date: _____ Period: _____

Atoms and Molecules Worksheet

Using the Periodic Table located inside your text book, answer the following questions.

Analyzing Data: The information below is from part of the Periodic Table of Elements. Use this information to answer the questions that follow.

IA							VIII
1 H 1							2 He 4
	IIA	IIIB	IVB	VB	VIB	VIIIB	
3 Li 7	4 Be 9	5 B 11	6 C 12	7 N 14	8 O 16	9 F 19	10 Ne 20
11 Na 23	12 Mg 24	13 Al 27	14 Si 28	15 P 31	16 S 32	17 Cl 35	18 Ar 40

1) How many protons, neutrons, and electrons are in each of the following atoms?

Lithium (Li)

p = _____

n = _____

e = _____

Magnesium (Mg)

p = _____

n = _____

e = _____

Carbon (C)

p = _____

n = _____

e = _____

Oxygen (O)

p = _____

n = _____

e = _____

Chlorine (Cl)

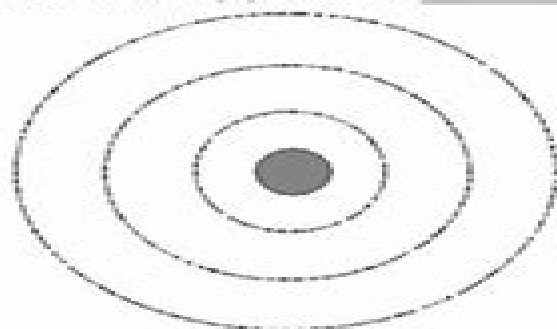
p = _____

n = _____

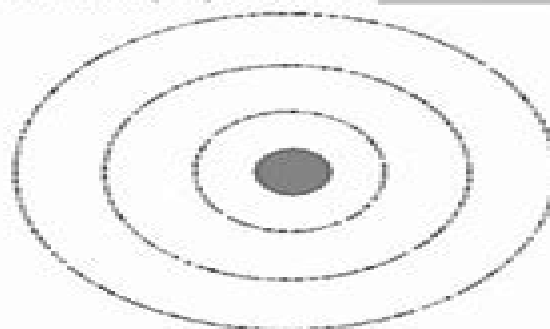
e = _____

2) Fill in the number of electrons in the correct energy levels below.

Number of Fluorine (F) electrons: _____



Number of Neon (Ne) electrons: _____



Is Fluorine Stable? _____

Explain your answer: _____

Is Neon Stable? _____

Explain your answer: _____

Worksheet 1 Rearranging Atoms

Dieter Mergel

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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Table of Contents Worksheet 1 Rearranging Atoms

1. Understanding the eBook Worksheet 1 Rearranging Atoms
 - The Rise of Digital Reading Worksheet 1 Rearranging Atoms
 - Advantages of eBooks Over Traditional Books
2. Identifying Worksheet 1 Rearranging Atoms
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Worksheet 1 Rearranging Atoms
 - User-Friendly Interface
4. Exploring eBook Recommendations from Worksheet 1 Rearranging Atoms
 - Personalized Recommendations
 - Worksheet 1 Rearranging Atoms User Reviews and Ratings
 - Worksheet 1 Rearranging Atoms and Bestseller Lists

5. Accessing Worksheet 1 Rearranging Atoms Free and Paid eBooks
 - Worksheet 1 Rearranging Atoms Public Domain eBooks
 - Worksheet 1 Rearranging Atoms eBook Subscription Services
 - Worksheet 1 Rearranging Atoms Budget-Friendly Options
6. Navigating Worksheet 1 Rearranging Atoms eBook Formats
 - ePub, PDF, MOBI, and More
 - Worksheet 1 Rearranging Atoms Compatibility with Devices
 - Worksheet 1 Rearranging Atoms Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Worksheet 1 Rearranging Atoms
 - Highlighting and Note-Taking Worksheet 1 Rearranging Atoms
 - Interactive Elements Worksheet 1 Rearranging Atoms
8. Staying Engaged with Worksheet 1 Rearranging Atoms
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Worksheet 1 Rearranging Atoms
9. Balancing eBooks and Physical Books Worksheet 1 Rearranging Atoms
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Worksheet 1 Rearranging Atoms
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Worksheet 1 Rearranging Atoms
 - Setting Reading Goals Worksheet 1 Rearranging Atoms
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Worksheet 1 Rearranging Atoms
 - Fact-Checking eBook Content of Worksheet 1 Rearranging Atoms
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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